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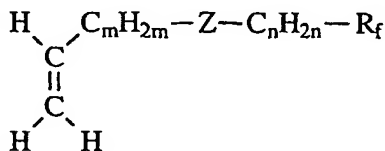
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(54) Title: FLUOROCHEMICAL TREATMENT FOR SILICON ARTICLES



(1)

(57) Abstract: Silicon substrates having Si-H bonds are chemically modified using a fluorinated olefin having the formula wherein m is an integer greater than or equal to 1; n is an integer greater than or equal to 0; Z is a divalent linking group; and R<sub>f</sub> is a highly fluorinated organic group.

## FLUOROCHEMICAL TREATMENT FOR SILICON ARTICLES

### TECHNICAL FIELD

5 The present invention relates generally to chemical modification of silicon substrates, and more specifically to chemical modification of silicon-based microelectromechanical systems (that is, MEMS) devices.

### BACKGROUND

10 MEMS devices are used, for example, as airbag accelerometers, microengines, optical switches, gyroscopic devices, sensors, and actuators. MEMS devices are typically manufactured from a silicon wafer using microlithography, and have freestanding active silicon elements (for example, gears, hinges, levers, slides, and mirrors) that must be free to move, rotate, etc. The inherent mechanical nature of MEMS devices brings about a high and ever increasing level of complexity to their production and reliability.

15 MEMS devices generally have large surface-area-to-volume ratios, and thus their performance is often dominated by strong adhesion (that is, stiction) of active silicon elements to one another and/or to the supporting silicon substrate. Stiction may be caused by capillary forces (for example, resulting from high critical surface tension of silicon surfaces of MEMS devices), electrostatic forces, van der Waals forces, and/or "chemical" forces such as hydrogen bonding and solid bridging. Stiction may occur, for example, 20 during release of MEMS elements from via etching, or at a later time (for example, during use).

25 Stiction, friction, and wear are major problems limiting both the production yield and useful lifetime of many MEMS devices, and have plagued the MEMS industry since its inception.

To address the problem of stiction, various treatments that reduce stiction (that is, anti-stiction treatments) are commonly employed. Such treatments typically reduce capillary forces by reducing the surface energy of MEMS surfaces (for example, resulting in a high advancing and/or receding contact angle with water). Ideally, anti-stiction 30 treatments are stable at typical operating temperatures of MEMS devices (for example, in the range of from -40 to +130 °C). In some cases, an anti-stiction treatment may also

serve to lubricate moving MEMS elements, thereby ensuring proper function and reducing the rate of wear.

Many anti-stiction and/or lubrication treatments are mere liquid coatings, and thus prone to loss or removal over time. Other anti-stiction treatments utilize various grafting techniques to chemically bond an anti-stiction coating to silicon surfaces of the MEMS device. However, and notwithstanding any reduction in stiction that may be achieved by such methods, there continues to be a need for further improvements in anti-stiction treatments.

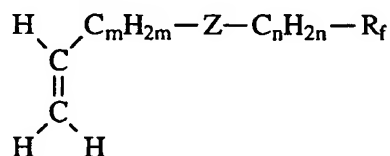
### SUMMARY OF THE PRESENT INVENTION

In one aspect, the present invention provides a method for modifying a silicon substrate comprising:

providing a silicon substrate having a plurality of Si-H bonds;

providing a composition comprising:

a fluorinated olefin having the formula:



wherein

m is an integer greater than or equal to 1,

n is an integer greater than or equal to 0,

Z is a divalent linking group,

R<sub>f</sub> is a highly fluorinated organic group, and

optional solvent; and

contacting the composition with the silicon substrate under conditions such that the fluorinated olefin becomes covalently attached to the surface of the silicon substrate.

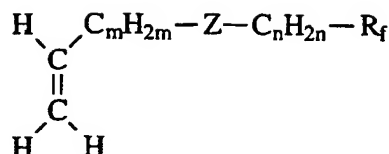
In another aspect, the present invention provides an article comprising a chemically modified silicon substrate prepared according to a method comprising:

providing a silicon substrate;

etching the silicon substrate to form a plurality of Si-H bonds;

providing a composition comprising:

a fluorinated olefin having the formula:



wherein

m is an integer greater than or equal to 1,

n is an integer greater than or equal to 0,

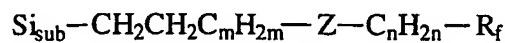
5 Z is a divalent linking group,

R<sub>f</sub> is a highly fluorinated organic group, and

optional solvent; and

contacting the composition with the silicon substrate under conditions such that the fluorinated olefin becomes covalently attached to the silicon substrate.

10 In another aspect, the present invention provides an article comprising a chemically modified silicon substrate, wherein the chemically modified substrate comprises at least one silicon atom that is covalently bonded to an organic group having the formula:



15 wherein

m is an integer greater than or equal to 1;

n is an integer greater than or equal to 0;

Z is a divalent linking group;

R<sub>f</sub> is a highly fluorinated organic group; and

20 Si<sub>sub</sub> is a substrate silicon atom directly bonded to at least one additional substrate silicon atom.

Surface-modified silicon substrates prepared according to the present invention exhibit outstanding properties (for example, hydrophobicity and/or oleophobicity). The methods and materials of the present invention have applicability to the fabrication of a broad range of silicon-based articles, especially regarding MEMS devices, where they provide useful anti-stiction coatings.

25

### DETAILED DESCRIPTION

The present invention concerns materials and methods for modifying a silicon substrate having chemically accessible Si-H bonds. Modification is achieved by hydrosilylation of the Si-H bonds with at least one fluorinated olefin. In hydrosilylation, a Si-H bond is added across a terminal carbon-carbon double bond of an olefin, typically attaching a silicon atom to the terminal carbon atom of the olefin. Modification of the silicon substrate typically occurs at its surface, although modification may also occur within the body of the silicon substrate in region(s) that are chemically accessible (for example, as in the case of a porous silicon substrate).

Silicon-hydride (that is, Si-H) bonds may be formed on a silicon substrate, for example, by treating an oxidized silicon surface with an etching agent. Suitable etching agents, acidic or basic, are widely known and may be used to prepare hydrogenated silicon surfaces, suitable for use in practicing the present invention, having up to 3 hydrides per silicon atom depending on the etching agent(s) used.

Exemplary acidic etching agents that result in various silicon hydrides on the silicon substrate surface include 40 percent by weight aqueous ammonium fluoride; aqueous hydrofluoric acid; and anhydrous hydrofluoric acid etch cleaning compositions such as those described, for example, in U. S. Pat. No. 6,310,018 (Behr et al.).

In an exemplary basic method, aqueous base (for example, sodium hydroxide) can be used as an etching agent to provide a hydrogenated silicon surface.

The etching duration, etching agent concentration, and temperature are generally interrelated. Typically, the etching agent concentration ranges from 27 molar to 12 molar, temperatures of the etching process range from 18 to 30 °C, and the duration of the etching ranges from 1 to 5 minutes (min), although other concentrations, temperatures, and etching durations may be used.

Subsequent to etching, the substrate is desirably handled under an inert atmosphere (for example, nitrogen or argon) in order to prevent conversion of the Si-H bonds to Si-O bonds.

Once a silicon substrate has been etched to form a surface having Si-H bonds, the fluorinated olefin, whether present as a mixture, in solution, or in pure form is applied to the silicon substrate such that the fluorinated olefin comes into contact with Si-H bonds (for example, on an exposed surface of the silicon substrate).

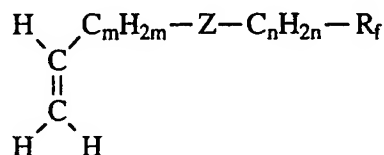
Useful silicon substrates include any form of solid silicon. Exemplary silicon substrates include polycrystalline silicon (that is, polysilicon), silicon nanocrystals, amorphous silicon, porous silicon, thin films of any of these forms of solid silicon, and/or combinations thereof. The silicon substrate may have any shape. Desirably, the silicon substrate is planar. More desirably, the substrate has the form of a wafer or a chip.

In some embodiments, the silicon substrate may comprise one or more microelectromechanical elements (for example, a MEMS device). Exemplary MEMS devices include accelerometers; pressure, mass flow, and yaw rate sensors; displays; optical scanners; adaptive optics; optical modulators; optical attenuators; dynamic gain equalizers; optical, microwave, and radio frequency switches; tunable capacitors and inductors; micro-relays; chemical sensors; and micro-valves.

MEMS devices may be fabricated, for example, by commercially known methods such as those available from JDS Uniphase MEMS Business Unit (Cronos), Research Triangle Park, NC; Sandia National Laboratories, Albuquerque, New Mexico; or Tronic's Microsystems, Grenoble, France. An exemplary assembly procedure is described in "MUMPs Design Handbook", revision 7.0 (2001), available from JDS Uniphase MEMS Business Unit (Cronos).

Methods and articles according to the present invention are useful for additional applications beyond MEMS device anti-stiction treatments (for example, stabilization of porous silicon, preparation of stable Si electrodes, silicon surface passivation, immobilization of DNA on Si, and in manufacture of microfluidic devices).

The fluorinated olefin, useful for practicing the present invention, includes at least one material having the formula:



wherein m, n, Z, and R<sub>f</sub> are defined as follows:

m is an integer greater than or equal to 1, desirably in the range of from 1 to 20, more desirably in the range of from 1 to 10. In some desired embodiments, m is 1.

Without wishing to be bound by theory, it is believed that the presence of one or more methylene groups (that is, m greater than or equal to 1) provides enhanced reactivity to the fluorinated olefin as compared with electron withdrawing groups such as perfluoroalkyl or

perfluoroalkylene, and provides enhanced chemical and environmental stability as compared with strong electron donating groups such as alkoxy, dialkylamino, and the like.

n is an integer greater than or equal to 0, desirably in the range of from 0 to 20, more desirably in the range of from 0 to 3.

5 Z is a covalent bond or a divalent linking group, including, for example, straight chain, branched chain or cyclic alkylene; arylene; straight chain, branched chain, or cyclic aralkylene; oxygen; carbonyl; sulfur; sulfonyl; amino (including substituted amino, for example, alkylamino); and combinations thereof such as, for example, sulfonamido, carboxamido, carbonyloxy, oxycarbonyl, alkyleneoxy, sulfonamidoalkylene, carboxamidoalkylene, or poly(oxyalkylene). Desirably, Z is a covalent bond, -O-, or

$\text{—}\overset{\text{O}}{\underset{\text{R}^1}{\text{N}}}\text{C—}$ , wherein R<sup>1</sup> is alkyl (for example, methyl, ethyl) or H. More desirably, Z is -

O- or  $\text{—}\overset{\text{O}}{\underset{\text{H}}{\text{N}}}\text{C—}$ .

R<sub>f</sub> is a highly fluorinated organic group. As used herein, the term "highly fluorinated organic group" means an organic group comprising at least 3 carbon atoms and containing fluorine in an amount greater than or equal to 40 percent by weight. Desirably, R<sub>f</sub> contains fluorine in an amount greater than or equal to 50 percent by weight, more desirably in an amount greater than or equal to 60 percent by weight. R<sub>f</sub> may be, for example, branched, straight chain (that is, not branched), and/or cyclic. R<sub>f</sub> may be, for example, a fluorine-substituted alkyl, alkaryl, aralkyl, or aryl group, optionally substituted with one or more additional groups including, for example, oxa, alkoxy, and allyloxy groups.

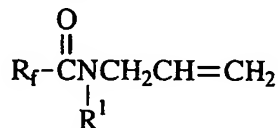
In some embodiments, R<sub>f</sub> may comprise an alkyl, alkylene, alkoxyalkyl group, and/or a derivative thereof that is at least partially fluorinated. In some embodiments, R<sub>f</sub> has from 4 to 30 carbon atoms, although other numbers of carbon atoms are also useful. Desirably, R<sub>f</sub> comprises one or more perfluoroalkyl and/or perfluoroalkylene groups. Exemplary perfluoroalkyl groups include trifluoromethyl, pentafluoroethyl, perfluoropropyl (for example, perfluoroisopropyl), perfluorobutyl (for example, perfluoro-

n-butyl, perfluoroisobutyl, perfluoro-tert-butyl), perfluorooctyl (for example, perfluoro-n-octyl, perfluoroisooctyl), and perfluorodecyl. Desirably, perfluoroalkyl groups have from 4 to 12 carbon atoms. Exemplary perfluoroalkylene groups include hexafluoropropylene, octafluorobutylene, and dodecafluorohexylene.

5 In some embodiments,  $R_f$  may comprise a poly(alkyleneoxy) group and/or partially or fully fluorinated derivative thereof. Exemplary perfluorinated polyalkyleneoxy groups include poly(tetrafluoroethyleneoxy) groups; mono- and divalent poly(hexafluoropropyleneoxy) groups; and groups consisting of block or randomly arranged difluoromethyleneoxy, tetrafluoroethyleneoxy, and hexafluoropropyleneoxy groups (for example, poly(tetrafluoroethyleneoxy-co-hexafluoropropyleneoxy) groups).

The fluorinated olefin, useful in practice of the present invention, may be obtained, for example, directly from a commercial supplier and/or prepared by known method(s), the choice typically depending on the specific nature of m, n, Z, and  $R_f$ .

15 In some embodiments of the present invention, the fluorinated olefin comprises at least one fluorinated allyl amide, desirably having the formula:



wherein  $R_f$  and  $R^1$  are as defined hereinabove. Such compounds may typically be prepared by reaction of a fluorinated ester, for example, having the formula  $R_f\text{C}(\text{O})\text{OCH}_3$ , with excess allylamine to form the corresponding allylamide in a manner as described, for example, in the Examples presented hereinbelow.

In some embodiments of the present invention, the fluorinated olefin comprises at least one fluorinated allyl ether, desirably having the formula:



or



wherein  $R_f$  is a highly fluorinated organic group as defined hereinabove. Such compounds may be prepared by allylating a fluorinated alcohol, for example, having the formula  $R_f\text{CH}_2\text{OH}$  or  $R_f\text{CH}_2\text{CH}_2\text{OH}$ . Typically, such fluorinated alcohols may be



allylated (for example, with excess allyl bromide) to form the corresponding allyl ethers in a manner as described, for example, in the Examples presented hereinbelow.

Typically, fluorinated alcohols (for example, having the formula  $R_fCH_2OH$  or  $R_fCH_2CH_2OH$ ) may be obtained directly from a commercial supplier and/or prepared, for example, by reduction of the corresponding carboxylic acid with lithium aluminum hydride or by catalytic hydrogenation over a copper-chromium oxide catalyst (for example, as described in U.S. Pat. No. 2,666,797 (Husted et al.)); by sodium borohydride reduction of the methyl ester of the corresponding fluorinated carboxylic acid (as described, for example, in U.S. Pat. No. 4,156,791 (Childs)); by sodium borohydride reduction of the corresponding fluorinated carboxylic acid halide (for example, as described in U.S. Pat. Nos. 3,293,306 (Bleu et al.) and 3,574,770 (Stump et al.)); by free-radical addition of methanol to perfluoroolefins (for example,  $CF_3CF=CF_2$  and  $C_5F_{11}CF=CF_2$ , as described, for example, by LaZerte and Kosher in J. Am. Chem. Soc., vol. 77, p. 910 (1955)); by free-radical telomerization of tetrafluoroethylene with methanol (for example, as described in U.S. Pat. No. 2,559,628 (Joyce)), and/or by direct fluorination of the corresponding acetate of a hydrocarbon alcohol (for example, as described in U.S. Pat. No. 5,488,142 (Fall et al.)), and subsequent hydrolysis.

Exemplary useful commercially available fluorinated alcohols include substituted and unsubstituted 1H,1H-dihydroperfluoroalkan-1-ols (for example, 2,2,3,3-tetrafluoro-1,4-butanediol; 1H,1H,3H-tetrafluoro-1-propanol; 1H,1H-pentafluoro-1-propanol; 2,2,3,3,4,4-hexafluoro-1,5-pentanediol; 1H,1H-heptafluoro-1-butanol; 1H,1H,6H,6H-octafluorohexanediol; 1H,1H,5H-octafluoro-1-pentanol; undecafluorocyclohexylmethanol; 1H,1H,7H-dodecafluoro-1-heptanol; 1H,1H-pentadecafluorooctan-1-ol; 1H,1H,9H-hexadecafluoro-1-nonanol; and 1H,1H,11H-eicosafluoro-1-undecanol); perfluoropolyetherdiols having a number average molecular weight of from 1,200 to 14,000 grams/mole (g/mole), such as, for example, those having the trade designation "FOMBLIN" (for example, "FOMBLIN Z DOL 2000", "FOMBLIN Z DOL 2500", "FOMBLIN Z DOL 4000", "FOMBLIN Z DOL TX", "FOMBLIN Z TETRAOL", "FOMBLIN AM 2001", "FOMBLIN AM 3001"), commercially available from Ausimont USA, Thorofare, New Jersey).

In some embodiments of the present invention, the fluorinated olefin desirably comprises a poly(perfluoroalkyleneoxy) group.

In some embodiments of the present invention, the fluorinated olefin desirably comprises a perfluoroalkanesulfonamido group such as, for example, a perfluorobutanesulfonamido or perfluorooctanesulfonamido group.

5 The fluorinated olefin may be present, for example, as a mixture (for example, a dispersion in solvent), in solution in solvent, or in pure form, and may optionally be present in combination with, for example, a hydrosilylation catalyst and/or a free-radical initiator as described hereinbelow.

Solvent that may, optionally, be used in conjunction with the fluorinated olefin, according to the present invention, may comprise one or more compounds that are liquid at 20 °C. Desirably, solvent should be inert (that is, substantially free of functional groups that would interfere with the hydrosilylation reaction). Exemplary suitable compounds that may constitute solvent include alkanes (for example, dodecane, hexadecane, isoundecane), hydrofluoroethers (for example, those marketed by 3M Company under the trade designations "3M NOVEC ENGINEERED FLUID HFE-7100", "3M NOVEC  
15 ENGINEERED FLUID HFE-7500", "3M NOVEC ENGINEERED FLUID HFE-7200"), chlorofluorocarbons (for example,  $\text{CF}_2\text{ClCFCl}_2$ ), and perfluorinated trialkylamines (for example, that marketed by 3M Company under the trade designation "3M FLUORINERT ELECTRONIC FLUID FC-70"). Desirably, optional solvent is chosen such that the fluorinated olefin and other optional components is/are soluble, at ambient temperature  
20 and at the concentrations utilized.

Suitable methods of applying the fluorinated olefin and any other optional components (for example, hydrosilylation catalyst, free-radical initiator, and/or solvent) to the silicon substrate include, for example, spraying, dip coating, wiping, and spin coating. In some embodiments of the present invention (for example, if the silicon substrate  
25 comprises a MEMS device), hydrosilylation is desirably carried out utilizing a solution of the fluorinated olefin in solvent. In such cases, the fluorinated olefin is typically present in an amount of from 0.05 to 20 percent by weight based on the total weight of solution, although other percentages are possible.

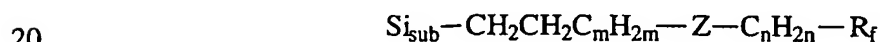
In some embodiments (for example, thermal hydrosilylation of a silicon substrate with a volatile fluorinated olefin), as an alternative to the aforementioned application  
30 methods, the fluorinated olefin in vapor form may be brought into contact with the silicon substrate under conditions wherein hydrosilylation can occur. Desirably, in such

embodiments the fluorinated olefin consists essentially of a single fluorinated olefin, although mixtures can be used. Such technique may be especially desirable, if utilizing a fluorinated olefin that is prone to secondary reactions (for example, polymerization) under hydrosilylation conditions.

5 After coating the silicon substrate, but prior to hydrosilylation, optional solvent that is present may, optionally, be removed (for example, by evaporation).

Typically, the fluorinated olefin is applied as a layer to at least a portion, desirably all, of the substrate surface to be treated. Desirably, the fluorinated olefin forms a monolayer (for example, a self-assembled monolayer) on the surface of the silicon  
10 substrate. As a surface treatment for a silicon substrate, the layer of the fluorinated olefin may be of any thickness, but for MEMS device applications, the layer of the fluorinated olefin, after reaction with the silicon substrate and removal of optional solvent, typically has a thickness in the range of from 0.5 to 10 nanometers (nm), desirably in the range of from 1 to 5 nm, more desirably in the range of from 1 to 2.5 nm.

15 Hydrosilylation of the fluorinated olefin with substrate Si-H groups (for example, Si<sub>sub</sub>-H, see below) typically, and desirably, results in a chemically modified silicon substrate, wherein the chemically modified substrate comprises at least one silicon atom that is covalently bonded to an organic group having the formula:



wherein

m, n, Z, and R<sub>f</sub> are as previously defined, and

Si<sub>sub</sub> represents a substrate silicon atom directly bound (for example, coordinated) to at least one and typically three additional substrate silicon atoms.

25 Desirably, hydrosilylation of the fluorinated olefin with Si<sub>sub</sub>-H groups results in chemical modification of the entire surface (for example, by formation of a layer of organic material) of the silicon substrate, although partial coverage can also be useful. Desirably, the thickness of the layer of organic material corresponds to the length of a single olefin molecule (that is, a monolayer), although thicker layers of organic material  
30 (for example, resulting from secondary reactions of the fluorinated olefin or other components) are also typically useful.

Hydrosilylation may proceed spontaneously or a catalyst, free-radical initiator (for example, free-radical photoinitiator), and/or energy may be utilized. Exemplary forms of energy include thermal energy and/or electromagnetic radiation (for example, having a wavelength in the range of 200 to 400 nm). Typically, heating a silicon substrate having surface Si-H bonds at temperatures up to 180 °C or more, in the presence of a fluorinated olefin (for example, fluorinated allyl ether), for a period of minutes to several hours (hr) (for example, 3 hr), or even longer, is an effective means of carrying out the hydrosilylation reaction. Under such conditions, a high boiling solvent (for example, having a boiling point of greater than or equal to 180 °C) may desirably be used with the fluorinated olefin. Further, under such conditions it is desirable that the fluorinated olefin is neither thermally unstable nor volatile at the temperatures utilized. Otherwise, it may decompose or evaporate before it can react with surface Si-H bonds of the silicon substrate.

While not required, in some embodiments of the invention at least one of a hydrosilylation catalyst and/or a free-radical initiator (for example, a free-radical photoinitiator) may be utilized to facilitate reaction of the fluorinated olefin with Si-H bonds. Desirably, the hydrosilylation catalyst and/or the free-radical initiator, if present, is/are in solution with the fluorinated olefin and any optional solvent.

Hydrosilylation catalyst may include, for example, one or more of a precious metal-containing hydrosilylation catalyst (for example, chloroplatinic acid, platinum deposited on a substrate, platinum complexed with an organic ligand (for example, alcohol, aldehyde), and/or a rhodium halide complex). In some embodiments, hydrosilylation catalyst can be a hydrosilylation photocatalyst (that is, a hydrosilylation catalyst precursor that is activated by exposure to electromagnetic radiation to form a hydrosilylation catalyst). Typically, useful electromagnetic radiation has at least one wavelength in the range of, for example, from 200 to 700 nm, although other wavelengths may be used. Useful photocatalysts include, for example: platinum azo complexes, as described, for example, in U.S. Pat. No. 4,670,531 (Eckberg); ( $\eta^4$ -cyclooctadiene)diarylplatinum complexes as described, for example, in U.S. Pat. No. 4,530,879 (Drahnak); platinum alkyne complexes as described, for example, in U.S. Pat. No. 4,603,215 (Chandra et al.); ( $\eta^5$ -cyclopentadienyl)trialkylplatinum complexes as described, for example, in U.S. Pat. Nos. 4,510,094 (Drahnak) and 4,916,169 (Boardman et al.);

platinum acetylacetonate complexes as described, for example, in U.S. Pat. No. 5,145,886 (Oxman et al.).

Hydrosilylation catalyst, if utilized, is typically present in an effective amount, for example, in an amount ranging from 1 to 1000 parts by weight of hydrosilylation catalyst per one million parts by weight of the fluorinated olefin, desirably from 10 to 200 parts by weight hydrosilylation catalyst, per one million parts of the fluorinated olefin, although other amounts and ranges can be utilized.

Free-radical initiator generates free radicals that can abstract H-atoms from silicon hydride bonds upon exposure to heat or electromagnetic radiation. Free-radical initiator may include at least one free-radical thermal initiator and/or at least one photoinitiator. Exemplary free-radical thermal initiators include organic peroxides (for example, benzoyl peroxide) and azo compounds (for example, 2,2'-azobisisobutyronitrile). Exemplary free-radical photoinitiators include one or more of benzoin and its derivatives (for example  $\alpha$ -methylbenzoin,  $\alpha$ -phenylbenzoin,  $\alpha$ -allylbenzoin,  $\alpha$ -benzylbenzoin), benzoin ethers such as benzil dimethyl ketal (available, for example, under the trade designation "IRGACURE 651", from Ciba Specialty Chemicals, Tarrytown, New York), benzoin methyl ether, benzoin ethyl ether, and benzoin n-butyl ether; acetophenone and its derivatives such as 2-hydroxy-2-methyl-1-phenyl-1-propanone (available, for example, under the trade designation "DAROCUR 1173" from Ciba Specialty Chemicals) and 1-hydroxycyclohexyl phenyl ketone (available, for example, under the trade designation "IRGACURE 184" from Ciba Specialty Chemicals), 2-methyl-1-[4-(methylthio)phenyl]-2-(4-morpholinyl)-1-propanone (available, for example, under the trade designation "IRGACURE 907" from Ciba Specialty Chemicals), and 2-benzyl-2-(dimethylamino)-1-[4-(4-morpholinyl)phenyl]-1-butanone (available, for example, under the trade designation "IRGACURE 369" from Ciba Specialty Chemicals); and combinations thereof. In some embodiments, free-radical photoinitiator may be combined with a platinum complex, for example, a platinum complex having one diolefin group that is eta-bonded to the platinum atom as described, for example, in U.S. Pat. No. 6,046,250 (Boardman et al.).

Following hydrosilylation, reaction components including optional solvent, unreacted olefin, and optional catalyst or free-radical initiator may be removed from the substrate (for example, by evaporation and/or rinsing, for example, with a volatile solvent

or liquid or supercritical CO<sub>2</sub>). Such removal of reaction components is highly desirable if the silicon substrate comprises a MEMS device.

The present invention will be more fully understood with reference to the following non-limiting examples in which all parts, percentages, ratios, and so forth, are by weight unless otherwise indicated.

### EXAMPLES

"Room temperature" in the following preparations and examples means in the range of from approximately 20 to 24 °C.

"Overnight" in the following preparations and examples means in the range of from approximately 14 to 16 hr.

Contact angles reported in Tables 1 and 2 were measured on the indicated surface modified silicon wafers using either deionized water ( $\geq 18$  megohms, using a purification system, obtained under the trade designation "MILLI-Q" from Millipore Corp., Bedford, MA) or anhydrous hexadecane, using a video contact angle measuring system (having the trade designation "VCA-2500XE", obtained from AST Products, Inc., Billerica, Massachusetts). The reported contact angles are an average value of contact angle measurements taken from opposite sides of three different drops. Typical uncertainty in the reported values of advancing and receding contact angles are  $\pm 2^\circ$  and  $\pm 4^\circ$ , respectively.

Unless otherwise noted, all reagents used in the following preparations and examples were obtained, or are available, from general chemical suppliers such as Aldrich Chemical Co., Milwaukee, Wisconsin, or may be synthesized by known methods.

#### Materials Used in the Examples

C<sub>2</sub>F<sub>5</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>CH<sub>2</sub>OH can be prepared as described in Example 3 of U.S. Pat. No. 5,437,812 (Janulis et al.).

"KRYTOX 157 FS(L)" is a trade designation for a fluorinated polyether having the formula CF<sub>3</sub>(CF<sub>2</sub>)<sub>2</sub>O[CF(CF<sub>3</sub>)CF<sub>2</sub>O]<sub>x</sub>CF(CF<sub>3</sub>)CO<sub>2</sub>H, and having a number average molecular weight M<sub>n</sub> of approximately 2,500 g/mole, obtained from E.I. du Pont de Nemours & Co., Wilmington, DE.

"FOMBLIN Z-DOL 2000" is a trade designation for fluorinated polyether having the formula  $\text{HOCH}_2\text{CF}_2[(\text{CF}_2\text{CF}_2\text{O})_q(\text{CF}_2\text{O})_p]\text{CF}_2\text{CH}_2\text{OH}$ , having a q to p ratio of 1, and having a number average molecular weight  $M_n$  of approximately 2,000 g/mole, obtained from Ausimont USA.

5            $\text{CF}_3(\text{CF}_2)_3\text{OC}_2\text{F}_4\text{OCF}_2\text{CH}_2\text{OC}_2\text{H}_4\text{OH}$  can be prepared, for example, as described in Example 23 of U.S. Pat. No. 5,437,812 (Janulis et al.).

$\text{CF}_3(\text{CF}_2)_7\text{CH}=\text{CH}_2$  (99% purity), obtained from Aldrich Chemical Co., was used after drying over 4A molecular sieves.

          1-Hexadecene (99% purity), obtained from ICN Biomedicals, Aurora, OH, was  
10       used after drying over 4A molecular sieves.

          1-Eicosene (>85% purity) was obtained from ICN Biomedicals, and used as received.

$\text{C}_7\text{F}_{15}\text{CH}_2\text{OH}$  (a mixture of approximately 70 percent linear and 30 percent  
branched isomers) was obtained from 3M Company, St. Paul, MN.

15            $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  was prepared according to the following method: To a mixture of 199.7 grams (g) of perfluoro-n-butyl iodide and 93.7 g of 10-undecen-1-ol in a mixture of 700 mL of acetonitrile and 300 mL of water was added a mixture of 53.8 g of sodium bicarbonate and 106.2 g of sodium dithionite in small portions with stirring. The reaction mixture was stirred at room temperature overnight and acidified with 1 N  
20       hydrochloric acid. The mixture was extracted with diethyl ether, and the combined organic phases were washed sequentially with saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride, then dried over anhydrous magnesium sulfate. Concentration afforded 234.4 g of crude  $\text{CF}_3(\text{CF}_2)_3\text{CH}_2\text{CHI}(\text{CH}_2)_8\text{CH}_2\text{OH}$  as a viscous, light amber liquid, which was used without further purification. To a slurry of  
25       130.0 g of zinc powder in 500 mL of ethanol was added 5.0 g of acetic acid. A solution of 230.0 g of  $\text{CF}_3(\text{CF}_2)_3\text{CH}_2\text{CHI}(\text{CH}_2)_8\text{CH}_2\text{OH}$  in 100 mL of ethanol was added dropwise with stirring over 1 hr, and the reaction mixture was heated at 50 °C for 4 hr. The mixture was filtered, the filtrate was concentrated to a viscous, light yellow liquid, and bulb-to-bulb distillation in several portions provided 97.3 g of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  as a  
30       colorless solid, b.p. 160-200 °C at 0.05 torr (7 Pa).

$\text{CF}_3(\text{CF}_2)_7(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  was prepared according to the following method: To a mixture of 41.10 g of perfluoro-n-octyl iodide and 11.92 g of 10-undecen-1-ol in a mixture of 100 mL of acetonitrile and 40 mL of water was added a mixture of 6.89 g of sodium bicarbonate and 13.58 g of sodium dithionite in small portions with stirring. The reaction mixture was stirred at room temperature overnight and acidified with 1 N hydrochloric acid. The mixture was extracted with diethyl ether, and the combined organic phases were washed sequentially with saturated aqueous sodium bicarbonate and saturated aqueous sodium chloride and dried over anhydrous magnesium sulfate. Concentration afforded 43.2 g of crude  $\text{CF}_3(\text{CF}_2)_7\text{CH}_2\text{CHI}(\text{CH}_2)_8\text{CH}_2\text{OH}$  as a white solid, which was used without further purification. To a slurry of 19.6 g of Zn powder in 150 mL of ethanol was added 4.0 g of acetic acid. A solution of the crude  $\text{CF}_3(\text{CF}_2)_7\text{CH}_2\text{CHI}(\text{CH}_2)_8\text{CH}_2\text{OH}$  prepared above in 50 mL of ethanol was added dropwise with stirring over 1 hr, and the reaction mixture was heated at 50 °C for 4 hr. The mixture was filtered, and the filtrate was concentrated to approximately 45 g of a soft, white solid. The crude  $\text{CF}_3(\text{CF}_2)_7(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  so obtained was used without further purification.

$\text{C}_4\text{F}_9\text{SO}_2\text{NHCH}_3$  was prepared according to the method described in Example 1 of PCT Publication No. WO 01/30873 (Savu et al.).

N-methylperfluorooctanesulfonamide ( $\text{C}_8\text{F}_{17}\text{SO}_2\text{NHCH}_3$ ) and N-ethylperfluorooctanesulfonamide ( $\text{C}_8\text{F}_{17}\text{SO}_2\text{NHCH}_3$ ) can be prepared generally according to the method described in Example 1 of U.S. Pat. No. 2,809,990 (Brown).

"SHFE" refers to a segregated hydrofluoroether available under the trade designation "3M NOVEC ENGINEERED FLUID HFE-7100", obtained from 3M Company.

"FSOLV" refers to perfluorinated solvent, obtained under the trade designation "3M FLUORINERT ELECTRONIC LIQUID FC-70", obtained from 3M Company.

#### Preparation of $\text{C}_7\text{F}_{15}\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$ (E1):

To a stirred solution of 124 g of  $\text{C}_7\text{F}_{15}\text{CH}_2\text{OH}$  and 62 g of allyl bromide in 1240 g of dimethyl sulfoxide was dropwise added a solution of 28.1 g of potassium hydroxide in 35 g of water. The mixture was allowed to stir for 4 hr, during which time it separated



into two immiscible liquid phases. The reaction mixture was poured into a separatory funnel, and the lower layer was isolated and washed 3 times with water to remove any remaining solvent. The crude product was dried over anhydrous magnesium sulfate, filtered, and vacuum distilled to yield 103 g of  $C_7F_{15}CH_2OCH_2CH=CH_2$  (E1), b.p. 88-89 °C at 30 torr (4 kilopascals (kPa)).

Preparation of  $C_2F_5OC_2F_4OCF_2CH_2OCH_2CH=CH_2$  (E2):

A one-liter (L), three-neck round bottom flask was equipped with magnetic stirring, ice bath, and connection to a nitrogen bubbler. The flask was charged with a mixture of 100.4 g of  $C_2F_5OC_2F_4OCF_2CH_2OH$  and 500 milliliters (mL) of dimethyl sulfoxide. While stirring this mixture, a solution of 29.1 g potassium hydroxide in 29.1 g of water was added, followed by 56.3 g of allyl bromide. After 10 min, the ice bath was removed, the mixture was allowed to warm to room temperature, and the mixture was stirred overnight. Stirring was stopped, and the reaction mixture separated into two liquid phases. Using a separatory funnel, the lower (product) layer was isolated. The upper phase was diluted to a volume of one liter with water, and the small amount of heavier liquid phase which separated was combined with the product layer. The product was washed four times with 200 mL portions of water, dried over anhydrous magnesium sulfate, filtered, and vacuum distilled to yield 83.3 g of  $C_2F_5OC_2F_4OCF_2CH_2OCH_2CH=CH_2$  (E2), b.p. 93-97 °C at 140 torr (18 kPa).

Preparation of  $CF_3(CF_2)_3OC_2F_4OCF_2CH_2OC_2H_4OCH_2CH=CH_2$  (E3):

A one-liter, three-neck round bottom flask was equipped with magnetic stirring and connection to a nitrogen bubbler. The flask was charged with a mixture of 50.2 g of  $CF_3(CF_2)_3OC_2F_4OCF_2CH_2OC_2H_4OH$  and 500 mL of dimethyl sulfoxide. The mixture was stirred rapidly while a solution of 7.8 g of potassium hydroxide in 7.4 g water was added, followed by 19.2 g of allyl bromide. The mixture was allowed to stir for 2.75 hr, at which time a solution of 0.91 g of potassium hydroxide in 0.97 g of water was added, and stirring continued overnight at room temperature. A solution of 1.1 g of potassium hydroxide in 1.1 g of water was then added, and stirring continued for another day at room temperature. At this point, the reaction mixture consisted of two liquid phases. The lower

(product) layer was isolated using a separatory funnel, washed three times with 150 mL portions of water, dried over anhydrous magnesium sulfate, filtered (the filter cake was washed with a few mL of 1,1,2-trichlorotrifluoroethane, and the washings were combined with the product), then volatile components were removed using a rotary evaporator at water aspirator pressure. Analysis of the product by gas chromatography showed residual alcohol starting material to still be present, so the product was recombined with 100 mL of dimethyl sulfoxide, 0.87 g of allyl bromide, and a solution of 0.4 g of potassium hydroxide in 0.56 g of water, and the mixture was stirred overnight at room temperature (approximately 20 °C). The lower liquid layer was isolated, then washed with water, dried over anhydrous magnesium sulfate, and filtered. Vacuum distillation gave 44.3 g of  $\text{CF}_3(\text{CF}_2)_3\text{OC}_2\text{F}_4\text{OCF}_2\text{CH}_2\text{OC}_2\text{H}_4\text{OCH}_2\text{CH}=\text{CH}_2$  (E3), b.p. 110-112 °C at 15 torr (2 kPa).

Preparation of  $\text{CH}_2=\text{CHCH}_2\text{OCH}_2\text{CF}_2[(\text{CF}_2\text{CF}_2\text{O})_q(\text{CF}_2\text{O})_p]\text{CF}_2\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E4):

To a stirred mixture of 100 g of glyme, 10.5 g of potassium hydroxide, 2 g of tetrabutylammonium bromide and 100 g of perfluoropolyetherdiol (FOMBLIN Z-DOL 2000) was added 22.5 g of allyl bromide over 1 hr at 45 °C. The mixture was heated at reflux (85 °C) for 15 hr. Solvents and excess allyl bromide were removed by distillation, the mixture was filtered to remove solids, and then water washed to obtain a fluorochemical lower phase. The fluorochemical product was heated at 95 °C under reduced pressure (15 torr (2 kPa)) to remove all volatiles, yielding 96 g of  $\text{CH}_2=\text{CHCH}_2\text{OCH}_2\text{CF}_2[(\text{CF}_2\text{CF}_2\text{O})_q(\text{CF}_2\text{O})_p]\text{CF}_2\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E4) having a number average molecular weight  $M_n$  of 2,050 g/mole as determined by fluorine nuclear magnetic resonance spectroscopy ( $^{19}\text{F}$  NMR).

Preparation of  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E5):

To a stirred solution containing 200 g of methanol and 200 g of  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CO}_2\text{H}$  (KRYTOX 157 FS(L)) was added 100 g of concentrated sulfuric acid. The mixture was heated at reflux (74 °C) for 4 hr. Water was added to obtain a lower fluorochemical phase, which was isolated. The isolated

fluorochemical phase was heated to 95 °C under a reduced pressure of 15 torr (2 kPa) to remove volatiles, yielding 184 g of  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CO}_2\text{CH}_3$  with a number average molecular weight  $M_n$  of 2,515 g/mole as determined by  $^{19}\text{F}$  NMR.

5  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CO}_2\text{CH}_3$  (166 g) was added over 1 hr to a stirred solution of 300 g of glyme and 28 g of sodium borohydride, and the mixture was heated at reflux (85 °C) for 15 hr. To this mixture was sequentially added 100 g of methanol, 200 g of water, and 67 g of concentrated sulfuric acid. Solvents were removed by distillation, and the mixture was filtered and washed with water to obtain a fluorochemical lower phase. The fluorochemical lower phase was heated at 95 °C and a  
10 reduced pressure of 15 torr (2 kPa) to remove volatiles, yielding 147 g of  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CH}_2\text{OH}$  with a number average molecular weight  $M_n$  of 2,485 g/mole determined by  $^{19}\text{F}$  NMR.

To a stirred solution of 100 g of glyme, 4.2 g of potassium hydroxide, 2 g of tetrabutylammonium bromide and 100 g of  
15  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CH}_2\text{OH}$  (prepared above) was added 9 g of allyl bromide over 1 hr with heating at 45 °C. The mixture was then heated at reflux (85 °C) for 15 hr. Solvents and excess allyl bromide were removed by distillation and the mixture was filtered and washed with water to obtain a fluorochemical lower phase. The fluorochemical product was heated at 95 °C and a reduced pressure of 15 torr (2 kPa) to  
20 remove volatiles yielding 96 g of  $\text{CF}_3(\text{CF}_2)_2\text{O}[\text{CF}(\text{CF}_3)\text{CF}_2\text{O}]_x\text{CF}(\text{CF}_3)\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E5) with a number average molecular weight  $M_n$  of 2,350 g/mole as determined by  $^{19}\text{F}$  NMR.

Preparation of  $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E6):

25 One kilogram of poly(ethylene glycol) methyl ether  $(\text{CH}_3\text{O}(\text{CH}_2\text{CH}_2\text{O})_w\text{CH}_2\text{CH}_2\text{OH})$  with number average molecular weight  $M_n$  of 550 g/mole, obtained from Aldrich Chemical Co., Milwaukee, WI, product No. 20,248-7) was reacted with excess acetic anhydride at 125 °C for 1 hr. The mixture was then heated to 160 °C under vacuum to remove excess acetic anhydride and acetic acid, yielding 1032 g  
30  $\text{CH}_3\text{O}(\text{CH}_2\text{CH}_2\text{O})_w\text{CH}_2\text{CH}_2\text{OC}(\text{O})\text{CH}_3$  having a number average molecular weight  $M_n$

of 590 g/mole as determined by proton nuclear magnetic resonance spectroscopy ( $^1\text{H}$  NMR).

A solution was prepared by diluting 1032 g of  $\text{CH}_3\text{O}(\text{CH}_2\text{CH}_2\text{O})_w\text{CH}_2\text{CH}_2\text{OC}(\text{O})\text{CH}_3$  with  $\text{CF}_2\text{ClCFCl}_2$  to a volume of 1800 mL.

5 This solution was fluorinated using 20 percent fluorine gas in nitrogen at 20 °C, by feeding the solution over 23 hr into a 10-L reactor containing 6,000 mL of  $\text{CF}_2\text{ClCFCl}_2$  and 4300 g of sodium fluoride and maintaining a 10 percent stoichiometric excess of fluorine gas throughout the reaction. After the reaction was completed, an excess of methanol was added. The solution was filtered to remove solids, washed with water to remove excess  
10 methanol, and  $\text{CF}_2\text{ClCFCl}_2$  solvent was distilled off under reduced pressure.

$\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CO}_2\text{CH}_3$  (2382 g) was obtained having a number average molecular weight  $M_n$  of 850 g/mole as determined by  $^{19}\text{F}$  NMR. To a stirred solution of 1500 g of glyme and 43 g of sodium borohydride was added 800 g of  $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CO}_2\text{CH}_3$  (0.94 mole, prepared above) over 1 hr and heated at  
15 reflux (85 °C) for 15 hr. To the reaction mixture was first added 400 g of methanol followed by 800 g of water and 280 g of concentrated sulfuric acid and heated to 95 °C to distill out solvents. The fluorochemical lower phase was heated to 95 °C at a pressure of 15 torr (2 kPa) to remove all volatiles, yielding 650 g of  $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CH}_2\text{OH}$  with a number average molecular weight  $M_n$  of 855 g/mole as determined by  $^{19}\text{F}$  NMR.

20 To a stirred solution of 85 g of  $\text{CF}_2\text{ClCFCl}_2$ , 30 g of glyme, 2.5 g of potassium hydroxide, 1 g of tetrabutylammonium bromide, and 30 g of  $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CH}_2\text{OH}$  (prepared above) was added 6.5 g of allyl bromide over 2 hr at 45 °C. Solvents and excess allyl bromide were removed by distillation, and 170 g of  $\text{CF}_2\text{ClCFCl}_2$  was added. The mixture was washed with 100 mL of 10 percent by weight  
25 hydrochloric acid, followed by washing with 2 x 100 g portions of water, then dried over anhydrous magnesium sulfate, and filtered. Volatiles were removed at 95 °C and a pressure of 15 torr (2 kPa), yielding 28 g of  $\text{CF}_3\text{O}(\text{CF}_2\text{CF}_2\text{O})_w\text{CF}_2\text{CH}_2\text{OCH}_2\text{CH}=\text{CH}_2$  (E6) with a number average molecular weight  $M_n$  of 948 g/mole as determined by  $^{19}\text{F}$  NMR.

30

Preparation of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_9\text{CH}=\text{CH}_2$  (E7):

To a mixture of 19.52 g of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  and 200 mL of 48 weight percent aqueous hydrobromic acid was slowly added 20 mL of concentrated sulfuric acid. The reaction mixture was heated at 100 °C for 24 hr, and poured into 1 liter of water. The mixture was extracted with hexanes, and the combined organic phases were washed with saturated aqueous sodium bicarbonate and dried over anhydrous magnesium sulfate. The solution was concentrated to an amber liquid, which was eluted through 3 inches of silica with hexanes. Concentration of the eluent yielded a light amber liquid, and bulb-to-bulb distillation gave 19.82 g of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_{10}\text{CH}_2\text{Br}$  as a clear, colorless liquid, b.p. 120-170 °C at 0.06 torr (8 Pa).

To a solution of 10.00 g of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_{10}\text{CH}_2\text{Br}$  in a mixture of 8 mL of dimethyl sulfoxide and 35 mL of tetrahydrofuran at -20 °C was added 6.5 g of potassium tert-butoxide in several portions. The reaction mixture was allowed to warm to 0 °C, and was poured into water. The mixture was extracted with diethyl ether, and the combined organic phases were washed with saturated aqueous sodium chloride, dried over anhydrous magnesium sulfate, and concentrated to a dark liquid. The crude product was eluted with hexanes through approximately 2.5 cm of silica, and the eluent was concentrated to a light amber liquid.

Distillation through a 6-inch (15 cm) Vigreux column yielded 4.97 g of  $\text{CF}_3(\text{CF}_2)_3(\text{CH}_2)_9\text{CH}=\text{CH}_2$  (E7) as a clear, colorless liquid, b.p. 50-52 °C at 0.05 torr (7 Pa).

Preparation of  $\text{CF}_3(\text{CF}_2)_7(\text{CH}_2)_9\text{CH}=\text{CH}_2$  (E8):

To a mixture of 29.0 g (0.049) of  $\text{CF}_3(\text{CF}_2)_7(\text{CH}_2)_{10}\text{CH}_2\text{OH}$  and 250 mL of 48 percent by weight aqueous hydrobromic acid was slowly added 25 mL of concentrated sulfuric acid. The reaction mixture was heated at 100 °C for 18 hr and poured into 1 liter of water. The mixture was extracted with hexanes, and the combined organic phases were washed with saturated aqueous sodium bicarbonate and dried over anhydrous magnesium sulfate. The solution was concentrated to a dark liquid, which was eluted through 3 inches (8 cm) of silica with hexanes. Concentration of the eluent yielded 20.2 g of

CF<sub>3</sub>(CF<sub>2</sub>)<sub>7</sub>(CH<sub>2</sub>)<sub>10</sub>CH<sub>2</sub>Br as a nearly white solid, which was used without further purification.

To a solution of 9.80 g of CF<sub>3</sub>(CF<sub>2</sub>)<sub>7</sub>(CH<sub>2</sub>)<sub>10</sub>CH<sub>2</sub>Br in a mixture of 8 mL of dimethyl sulfoxide and 40 mL of tetrahydrofuran at -10 °C was added 11.2 g of potassium tert-butoxide in several portions. The reaction mixture was allowed to warm to 0 °C and poured into cold water. The mixture was extracted with diethyl ether, and the combined organic phases were washed with saturated aqueous sodium chloride, dried over anhydrous magnesium sulfate, and concentrated to a dark semi-solid. The crude product was eluted through one inch (3 cm) of silica with hexanes, and the eluent was concentrated to a dark solid. Bulb-to-bulb distillation afforded 5.10 g of CF<sub>3</sub>(CF<sub>2</sub>)<sub>7</sub>(CH<sub>2</sub>)<sub>9</sub>CH=CH<sub>2</sub> as a colorless solid, b.p. 110-120 °C at 0.10 torr (13 Pa).

Preparation of C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>N(CH<sub>3</sub>)CH<sub>2</sub>CH=CH<sub>2</sub> (E9):

A 250 mL three-neck round-bottom flask equipped with thermometer, magnetic stirring, and addition funnel was charged with a solution of 75 g of C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>NHCH<sub>3</sub> in 80 g of dimethyl sulfoxide. The solution was stirred while a solution of 17.1 g potassium hydroxide in 20 g of water was added in 2 mL aliquots via pipette. The solution turned cloudy during the addition, then cleared with additional stirring. Allyl bromide (31.9 g) was charged to the addition funnel, and added dropwise to the reaction mixture. A rise in temperature of the reaction mixture to 75 °C was noted during the addition. The reaction mixture turned cloudy shortly after the addition was begun, and stirring was continued overnight at room temperature. The reaction mixture was poured into a separatory funnel and allowed to stand until the liquid layers had separated. A few mL water were added to the mixture to facilitate phase separation. The lower layer was drawn off, washed with water and brine, and dried over anhydrous magnesium sulfate, leaving 76.0 g of a clear, colorless crude liquid product which was vacuum distilled to yield 68.7 g C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>N(CH<sub>3</sub>)CH<sub>2</sub>CH=CH<sub>2</sub>, b.p. 97-101 °C at 19 torr (2.5 kPa).

Preparation of CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>C(O)NHCH<sub>2</sub>CH=CH<sub>2</sub> (E10):

To a 5-L flask containing 2068 g of triethylene glycol monomethyl ether (MTEG) in 500 mL dichloromethane was added 970 mL acetyl chloride over 4 hr. After addition

was complete, infrared spectroscopy showed a slight excess of acetyl chloride. The resultant product (MTEG acetate) was stripped of volatiles. Direct fluorination was done in a tubular reactor as described in U.S. Pat. No. 5,578,278 (Fall et al.), using perfluoro-N-methylmorpholine (PNMM) (available under the trade designation "PF-5052" from 3M Company) as the liquid medium. MTEG acetate (2647 g) was pumped into the reaction zone at 44.6 g/hr while maintaining a gas flow of 2,500 mL/min F<sub>2</sub> in 10,000 mL/min N<sub>2</sub>. At the end of the reaction (as judged by the presence of unreacted fluorine), addition of F<sub>2</sub> was stopped, then the reactor was flushed with N<sub>2</sub>, and 1300 g methanol was added over approximately 2.5 hr. The mixture was stirred approximately 0.5 hr longer. The product mixture was drained to an approximately 60-L separatory funnel and washed twice with about 15-20L water introduced from a bottom valve. The fluorochemical layer was then isolated and dried over about 450 g anhydrous magnesium sulfate, filtered, and most of the PNMM was removed using a 5-plate distillation apparatus. The residue (5310 g) was distilled at about 740 torr (101 kPa) on a 30 cm Vigreux column to give 1256 g of CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>CO<sub>2</sub>CH<sub>3</sub>, b.p. 140-150 °C. The above procedure was repeated yielding 2323 g of CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>CO<sub>2</sub>CH<sub>3</sub>, which was combined with the 1256 g previously obtained above to give 3579 g of CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>CO<sub>2</sub>CH<sub>3</sub> in 84 percent purity by gas chromatography (remainder inert fluorochemical).

A 250 mL round-bottomed flask was charged with 46.2 g CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>CO<sub>2</sub>CH<sub>3</sub> in 50 mL of methanol. Allylamine (7.8 g) was added in two approximately equal portions. After 15 min, gas chromatographic analysis showed complete conversion of the ester to the corresponding allyl amide. Solvent and excess allyl amine was removed by evaporation at reduced pressure. The remainder was distilled (b.p. 90-95 °C at 1 torr (133 Pa)) to obtain 41.3 g of cloudy product. This was filtered through a microfiber filter available under the trade designation "WHATMAN GRADE GF/B-BINDER-FREE GLASS FIBER" (1.0 micrometer pore size) from Whatman, PLC, 20/20 Maidstone, Kent, United Kingdom, and redistilled yielding 39 g of clear CF<sub>3</sub>OC<sub>2</sub>F<sub>4</sub>OC<sub>2</sub>F<sub>4</sub>OCF<sub>2</sub>C(O)NHCH<sub>2</sub>CH=CH<sub>2</sub> in 96 percent purity.

C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>N(CH<sub>3</sub>)(CH<sub>2</sub>)<sub>9</sub>CH=CH<sub>2</sub> (E11):

A 500 mL three-necked round-bottom flask equipped with magnetic stirring, oil bath heat, and connection to a nitrogen bubbler was charged with 17.2 g of N-methylperfluorobutanesulfonamide (C<sub>4</sub>F<sub>9</sub>SO<sub>2</sub>NHCH<sub>3</sub>) and 250 mL dimethyl sulfoxide.

5 Stirring was begun, and the solid dissolved within a few minutes. A solution of 4.0 g of potassium hydroxide in 5 mL of water was added via pipette, and stirring was continued for another 30 min while the oil bath was heated to 80 °C. By the end of this period most of the potassium hydroxide had dissolved. 11-Bromo-1-undecene, obtained from Aldrich Chemical Co. (95 percent purity, 12.3 g) was then added to the flask via pipette, and the

10 reaction mixture was left to stir for 19 hr at 80 °C. The oil bath was then removed to allow the mixture to cool, and the flask contents were poured into a separatory funnel. The mixture separated into two liquid phases, and the lower layer was drawn off. The upper phase was diluted with an equal volume of water, causing separation of a small amount of lower phase. This was also drawn off and combined with the first batch. The combined

15 crude product (18.0 g) was dissolved in 150 ml dichloromethane and washed successively with 100 ml portions of water, 10 weight percent aqueous potassium hydroxide solution, water, and brine. The dichloromethane solution was dried over anhydrous magnesium sulfate and filtered, then solvent was removed on a rotary evaporator at water aspirator pressure. This left 17.9 g light straw-colored liquid product. Bulb-to-bulb vacuum

20 distillation of 17.3 g crude product at 160-180 °C and 0.3 torr (4 Pa) gave 16.2 g of N-(10-undecenyl)-N-methylperfluorobutanesulfonamide as a clear, colorless liquid distillate.

Preparation of C<sub>8</sub>F<sub>17</sub>SO<sub>2</sub>N(CH<sub>3</sub>)CH<sub>2</sub>CH=CH<sub>2</sub> (E12)

A one-liter three-neck flask equipped with magnetic stirring and an addition funnel

25 was charged with 150 g of N-methylperfluorooctanesulfonamide and 570 g of dimethyl sulfoxide, and the mixture was stirred for a few minutes until the solid dissolved. To this stirred solution was added a solution of 25 g of potassium hydroxide in 25 g of water. This mixture was allowed to stir for 30 min, then 46 g of allyl bromide was added dropwise via the addition funnel. An exotherm was noted after a few minutes. The

30 reaction mixture was allowed to stir for 5.5 hr, during which time a white solid precipitated from solution. This solid was isolated by filtration, washed with 350 mL



dimethyl sulfoxide, and dissolved in 450 g of 1,1,2-trichlorotrifluoroethane. This solution was transferred to a separatory funnel, washed with four 200 mL portions of water, and dried over anhydrous sodium sulfate. After removal of solvent by distillation at atmospheric pressure, the product was vacuum distilled to yield 138 g of

5  $\text{C}_8\text{F}_{17}\text{SO}_2\text{N}(\text{CH}_3)\text{CH}_2\text{CH}=\text{CH}_2$ , b.p. 98 °C at 0.4 torr (50 Pa) to 105°C at 0.7 torr (90 Pa). The product was a clear, almost colorless liquid which solidified to a white, waxy solid upon standing at room temperature.

Preparation of  $\text{C}_8\text{F}_{17}\text{SO}_2\text{N}(\text{CH}_2\text{CH}_3)\text{CH}_2\text{CH}=\text{CH}_2$  (E13)

10 A two-liter three-neck flask equipped with magnetic stirring, addition funnel, and nitrogen blanket was charged with 250 g of N-ethylperfluorooctanesulfonamide and 960 g dimethyl sulfoxide. The mixture was stirred at room temperature until the solid had dissolved. Then, a solution of 35 g of potassium hydroxide in 38 g of water was added gradually over a period of about 1 min. The mixture was allowed to stir for several more

15 minutes, then the addition funnel was charged with 71 g of allyl bromide. The allyl bromide was added dropwise over a period of about 30 min, then the mixture was allowed to stir at room temperature overnight. The reaction mixture, which now consisted of two liquid phases, was poured into a separatory funnel and the lower layer was drawn off, washed twice with dimethyl sulfoxide, twice with water, and twice with brine. Vacuum

20 distillation yielded 216 g of  $\text{C}_8\text{F}_{17}\text{SO}_2\text{N}(\text{CH}_2\text{CH}_3)\text{CH}_2\text{CH}=\text{CH}_2$  as a clear, almost colorless liquid, b.p. 80-85 °C at 0.1 torr (10 Pa).

General Procedure A: Preparation of a Silicon Substrate having Surface Si-H Bonds

25 Silicon wafers (Si(111), 10 centimeter diameter disks of 1 millimeter thickness), obtained from WaferNet, Inc., San Jose, California, were cleaved into small pieces (approximately 1 cm x 3 cm). The pieces were cleaned by sequentially rinsing them in heptane, acetone, and isopropyl alcohol. The cleaned pieces were subjected for 10 min to ultraviolet light and ozone in an apparatus in which an ultraviolet lamp (5 inch by 5 inch square (12.5 cm by 12.5 cm) ultraviolet lamp, obtained from BHK, Inc. under the trade

30 designation "UV GRID LAMP", Claremont, California, Model No. 88-9102-02) was encased in a small sheet metal box (13 cm wide x 14 cm deep x 15 cm high) such that the lamp was suspended 8 cm above the bottom of the box. A small lab jack was used to

position silicon wafer pieces to be cleaned as close as possible to the ultraviolet lamp without physically contacting the lamp. The front of the box was a door, hinged at the top, which allowed samples to be inserted and removed. A small hole in one side of the box was attached to a source of oxygen that flowed into the box at a rate of approximately 1 to 5 standard liters per minute. Ozone was produced in situ by the action of ultraviolet light on molecular oxygen.

The wafer pieces were then individually immersed for 4 min in a 40 percent by weight solution of ammonium fluoride in water that had been sparged with nitrogen for at least 30 min to remove dissolved oxygen. This procedure etches away the native oxide (SiO<sub>2</sub>) on the wafer resulting in a hydride-terminated, nearly atomically flat Si(111) surface.

Finally, the silicon wafer pieces were washed with water and isopropyl alcohol and blown dry with nitrogen.

#### General Procedure B: Grafting an Olefin to a Silicon Substrate having Surface Si-H Bonds

A silicon wafer piece having surface Si-H bonds, prepared according to the General Procedure A (above), was placed in a test tube containing a 5 percent by volume solution of the desired olefin in either hexadecane or FSOLV, which had previously been sparged with nitrogen for 30 min to remove residual oxygen.

After introduction of the olefin, the test tube was sealed with a silicone rubber septum and the solution was sparged with nitrogen for an additional 15 min. The test tube was placed in a heat bath set so that the internal test tube temperature reached 200 °C. The heat bath consisted of a 1200 mL stainless steel beaker, obtained from Polar Ware, Sheboygan, WI, in a heating mantle specifically for metal beakers, obtained from Glas-Col, Catalog No. TM630, Terre Haute, IN. The stainless steel beaker was filled to within 3 cm of the top with chrome steel balls, approximately 3.2 mm in diameter (obtained from Hartford Ball Co., Hartford, Connecticut). The heat bath temperature was controlled using a digital temperature controller (Catalog No. 12113-50), obtained from Ace Glass, Vineland, New Jersey, and equipped with a stainless steel sheathed type K thermocouple (4.8 mm diameter, 10 cm length), which was placed in the heat bath. After 3 hr, the test tube was removed from the heat bath, and cooled to room temperature. The silicon wafer piece was removed from the test tube. If hexadecane was used as solvent, the silicon

wafer piece was then washed sequentially with heptane, acetone, and isopropyl alcohol. If FSOLV was used as solvent, the silicon wafer piece was then rinsed with SHFE. The silicon wafer piece was further rinsed sequentially with heptane, acetone, and isopropyl alcohol. The rinsed silicon wafer piece was then dried using a stream of dry nitrogen.

5

General Procedure C: Grafting an Olefin to a Silicon Substrate having Surface Si-H Bonds

A solution of 5 percent by weight of olefin, 5 percent by weight of 2-hydroxy-2-methyl-1-phenyl-1-propanone (obtained under the trade designation "DAROCUR 1173" from Ciba Specialty Chemicals) free-radical photoinitiator, and 90 percent by weight solvent was placed in a quartz tube and sparged with nitrogen for 15 min. A silicon wafer piece having surface Si-H bonds, prepared according to the General Procedure A (above), was then added and the tube was sealed under dry nitrogen. The solution, under dry nitrogen, was irradiated for 30 min using an ultraviolet fluorescent light (obtained from Osram Sylvania, Danvers, Massachusetts, under the trade designation "BLACKLIGHT BLUE BLB 15W"). The solution was positioned at a distance of about 2 cm from the light source. Wafers functionalized with non-fluorinated olefins were then sequentially washed in heptane, acetone, and isopropyl alcohol, then blown dry with dry nitrogen. Silicon wafers functionalized with fluorinated olefins were sequentially washed with SHFE, heptane, acetone, and isopropyl alcohol, then blown dry with dry nitrogen.

20

General Procedure D: Grafting an Olefin to a Silicon Substrate having Surface Si-H Bonds

A silicon wafer piece having surface Si-H bonds, prepared according to General Procedure A (above), was placed in a test tube containing a 5 percent by volume solution of the olefin indicated in Table 3 (below) in FSOLV, which had previously been sparged with nitrogen for 30 min to remove residual oxygen.

25

After introduction of the silicon wafer piece, the test tube was sealed with a silicone rubber septum and the solution was sparged with nitrogen for 15 more min. The test tube, still under nitrogen, was placed in a silicone oil heat bath set at 180 °C. The heat bath was obtained under the trade designation "INSTATHERM OIL BATH, HIGH FORM" from Ace Glass (Part No. 9603-02) filled with high temperature silicone oil (Ace Glass, Part No. 14115-12). Temperature was controlled using a temperature controller, obtained under the trade designation "THERM-O-WATCH L7-1100SA/28T", from

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Instruments for Research and Industry, Inc., Cheltenham, Pennsylvania. After 15 min in the oil bath, the test tube was removed, and cooled to room temperature by placing the test tube in room temperature water. The wafer piece was removed from the test tube. The wafer piece was rinsed with SHFE. The wafer piece was further rinsed sequentially with heptane, acetone, and isopropyl alcohol. The wafer piece was then blown dry with dry nitrogen.

### EXAMPLES 1-11 and COMPARATIVE EXAMPLES 1-3

A series of olefins corresponding to Examples 1-11 and Comparative Examples 2-3, as indicated in Table 1 (below), were grafted to silicon (111) wafer pieces having Si-H bonds according to General Procedure B (above). For Examples 1-10 and Comparative Examples 2 – 3, Table 1 reports advancing and receding contact angles of water and hexadecane with an olefin treated surface of the silicon wafer piece. Comparative Example 1 was a silicon chip, prepared according to the General Procedure A (above), but not treated with any olefin. The contact angles for water and hexadecane reported in Table 1 were measured using an etched silicon wafer piece surface having no subsequent olefin treatment.

TABLE 1

Example	Olefin	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
Comparative Example 1	none	none	89/69	<10/<10
Comparative Example 2	$\text{CF}_3(\text{CF}_2)_7\text{CH}=\text{CH}_2$	FSOLV	115/107	65/56
Comparative Example 3	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	hexadecane	110/103	27/26
1	E1	FSOLV	117/103	69/62
2	E2	FSOLV	118/111	71/63
3	E3	FSOLV	113/97	63/56

Example	Olefin	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
4	E4	FSOLV	123/92	69/49
5	E5	FSOLV	121/109	71/66
6	E6	FSOLV	125/97	72/52
7	E7	hexadecane	113/104	53/49
8	E8	FSOLV	121/107	67/61
9	E9	FSOLV	85/47	34/26
10	E10	FSOLV	113/59	61/44
11	E11	hexadecane	106/86	47/38

### EXAMPLE 12

A 0.5 cm x 0.5 cm square silicon chip having a series of unreleased (that is, still having a protective photoresist) MEMS cantilever beams (similar to that described in R. Maboudian et al., Tribology Letters, vol. 3, pp. 215-221 (1997)) was made utilizing the Cronos MUMPS process with the following noted changes. The unreleased beams were made of polycrystalline silicon and positioned parallel to the polycrystalline silicon surface at a distance of 2 micrometers. The beams were regularly spaced to form an array (20 micrometers pitch), 3.5 micrometers thick, 10 micrometers wide, and had lengths varying in intervals of 50 micrometers from 200 to 1500 micrometers. The chip also had reference posts having a height of 5.5 micrometers thickness.

To remove protective photoresist resulting from the MUMPS process, the chip was sonicated in acetone for 20 min using an ultrasonic bath, obtained under the trade designation "FISHER SCIENTIFIC FS3 ULTRASONIC CLEANER" (Fisher Scientific, Pittsburgh, Pennsylvania), then sonicated in isopropyl alcohol (as before) for 5 min.

Release of the MEMS cantilever beams and rinsing of the MEMS chips was performed in a "fill / drain" apparatus similar to that described in R. Maboudian et al., Sensors and Actuators A (2000), vol. 82, pages 219-223. The "fill / drain" apparatus consisted of a block of polytetrafluoroethylene into which had been drilled a large hole (3.8 cm deep and 3.8 cm diameter), forming a reservoir. A drain hole, drilled in the center

of the bottom of the reservoir, was fitted with a polytetrafluoroethylene stopcock assembly, allowing the reservoir to be filled when the stopcock was closed and drained when it was opened. Six small slots (9.5 mm diameter, 1.6 mm deep) were machined into the bottom of the reservoir in a circle around the drain (the drain hole was at the same level as the tops of the slots) to allow MEMS chips to be positioned under the surface of the liquid in the reservoir even after it had been drained.

The chip was placed in a slot of the "fill / drain" apparatus. CMOS grade 49 percent by weight aqueous HF (J.T. Baker, Phillipsburg, New Jersey) was added and allowed to sit for 3 min to release the cantilever beam MEMS devices. The apparatus was then drained. A volume of 25 mL deionized water ( $\geq 18$  megohms) from a purification system, obtained under the trade designation "MILLI-Q" was then added to the apparatus to displace the aqueous HF. The water was then drained, and the process was repeated nine additional times, such that total water washing lasted for 10 min, and a total of 250 mL water was used. After the last of the water was drained, 5 mL of CMOS grade 49 percent by weight aqueous HF was added and allowed to etch the MEMS chip again for 3 min. The aqueous HF was again drained. The etched chip was then sequentially washed in isopropyl alcohol for 10 min, and SHFE for 10 min, each wash being performed in the same manner as the water wash above, with 10 fill and drain cycles over 10 min. Washing with FSOLV was accomplished by adding 5 mL of FSOLV and draining, repeating this two times for a total of three fill and drain cycles.

The chip was removed from the "fill / drain" apparatus and placed in a test tube containing a 5 percent by weight solution of E1 in FSOLV. This solution had been sparged 15 min with nitrogen before use. The solution was heated at 200 °C for 1 hr, and allowed to cool. After cooling, the MEMS chip was removed from the E1 and FSOLV solution, placed in a small glass cup (approximately 2 ml) containing SHFE and washed by adding 30 mL additional SHFE to the small cup over 10 min time. Drying the MEMS chip was accomplished by placing the small glass cup containing SHFE and the MEMS chip in an oven at 140 °C for 10 min.

To determine whether the MEMS cantilever beams were adhered to the silicon chip surface or not, the MEMS chip was examined using an optical microscope (obtained under the trade designation "OPTIPHOT 150" from Nikon, Nelville, New York). The MEMS cantilever beams were examined under high magnification (1000x) and the

reference posts were brought into focus. Beams having a free tip that was in the same focal plane as the reference posts were judged to be released and free of stiction. Beams that were not simultaneously in focus (with the reference posts) at their free tip, and had a focal plane closer to the silicon chip surface than the top surface of the reference post were judged to be stuck to the silicon chip surface (that is, exhibited stiction).

Treated MEMS cantilever beams prepared above, up to at least 1500 micrometers in length, were observed to be free from stiction to the silicon chip surface by this technique.

#### COMPARATIVE EXAMPLE 4

A MEMS chip containing cantilever beams as in Example 12 was sonicated in acetone for 20 min, then isopropyl alcohol for 5 min to remove photoresist. The chip was then placed in a slot of the "fill / drain" apparatus of Example 11 and etched with 5 mL 49 percent by weight aqueous HF for 3 min, then washed with 10 x 25 mL portions of deionized (>18 megohms) water, and then further washed with 10 x 25 mL portions of isopropyl alcohol. The etched chip was removed, placed in a small glass cup containing isopropyl alcohol, and the cup, MEMS chip and isopropyl alcohol were then dried in an oven at 140 °C for 10 min. By this method, using the analysis method of Example 12, cantilever beams of less than or equal to 400 micrometers in length released properly and did not show evidence of stiction. However, beams longer than 400 micrometers did not release properly from the silicon chip surface (that is, they were stuck to the chip surface).

#### EXAMPLES 13-21 AND COMPARATIVE EXAMPLES 5-13

Examples 13-16, and Comparative Examples 5, 6, and 8, were prepared by General Method C.

Example 17, was carried out by General Method C with the following changes: The solution consisted of 10 parts by weight of 50 percent by weight tert-butyl peroxy-2-ethylhexanoate in odorless mineral spirits (obtained under the trade designation "TRIGONOX 21-C50" from Akzo Nobel, Chicago, Illinois) as initiator, 10 parts by weight of E1, and 80 parts by weight SHFE solvent. A germicidal light emitting primarily 254 nm electromagnetic radiation (obtained under the trade designation "15 W G15T8" from Osram Sylvania) was used in place of the BLACKLIGHT BLUE BLB 15W bulb.

Example 18 was carried out as in Example 17 except that no exposure to light was used. Instead, the sample was warmed in an oil bath at 50 °C for 30 min.

Example 19 was carried out according to General Method C, except that a polycrystalline silicon wafer (10,000 Å undoped polysilicon on a 100 mm diameter Si(100) wafer obtained from WaferNet; prepared according to General Procedure A with the exception that the wafer piece was etched in CMOS grade 49 percent by weight aqueous HF for 3 min instead of etching in the ammonium fluoride solution) was used as the substrate, and E1 was used at 45 weight percent in a mixture of 5 weight percent 2-hydroxy-2-methyl-1-phenyl-1-propanone and 50 weight percent heptane.

Example 20 was carried out as in Example 19 except that E1 was used at 5 weight percent in a mixture of 5 weight percent 2-hydroxy-2-methyl-1-phenyl-1-propanone and 90 weight percent heptane.

Example 21 was carried out using a 1 x 1 cm piece of silicon (100) (obtained from WaferNet). The piece was cleaned by sequentially rinsing it in heptane, acetone, and isopropyl alcohol. The cleaned piece was subjected for 10 min to ultraviolet light and ozone using the ozone apparatus described in General Procedure A. The silicon piece was placed in a test tube. Acetone was added and the mixture was sonicated for 20 min. The acetone was decanted. Isopropyl alcohol was added and the mixture was sonicated for 20 min. The isopropyl alcohol was decanted and the silicon piece was allowed to dry in air.

The silicon piece was placed into CMOS-grade 49 weight percent aqueous HF and allowed to stand for 3 min. The silicon piece was removed from the HF and placed in approximately 100 mL of water ( $\geq 18$  megohm resistivity) in a plastic container. The silicon piece was again placed into CMOS-grade 49 weight percent aqueous HF and allowed to stand for 3 min. The silicon piece was removed from the HF, and then placed in approximately 100 mL isopropyl alcohol in a plastic container. It was then removed from the isopropyl alcohol and placed in small glass cup (approximately 2 mL volume) filled with isopropyl alcohol. The isopropyl alcohol was displaced with 30 mL of heptane over 10 min. The silicon piece was placed into a quartz tube containing a solution of 45 weight percent E1 and 5 weight percent 2-hydroxy-2-methyl-1-phenyl-1-propanone in heptane that had been sparged with nitrogen for 15 min. The silicon piece was irradiated for 30 min as in General Procedure B. The silicon piece was removed from the tube and placed into a small glass cup (approximately 2 mL volume) filled with heptane. The



silicon piece was rinsed by adding heptane (30 mL) to the cup over 10 min, followed by 30 ml isopropyl alcohol added over 10 min, and 30 ml of SHFE over 10 min. The wafer was removed from the cup and blown dry with nitrogen.

Comparative Examples 7 and 9 were carried out according to General Method C except that no solvent was used. The solution consisted of 5 percent by weight of 2-hydroxy-2-methyl-1-phenyl-1-propanone in the indicated olefin.

Comparative Examples 10 and 11 were carried out as in Comparative Examples 7 and 6, respectively, except that 2,2-dimethoxy-2-phenylacetophenone was used in place of 2-hydroxy-2-methyl-1-phenyl-1-propanone.

Comparative Examples 12 and 13 were carried out as in Comparative Examples 10 and 11, respectively, except that five freeze-thaw cycles were used to deoxygenate the tube contents.

For Examples 13-21 and Comparative Examples 5-13, the olefin and solvent utilized and resulting contact angle measurements are reported in Table 2 (below).

TABLE 2

Example	Olefin	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
Comparative Example 5	$\text{CF}_3(\text{CF}_2)_3\text{CH}=\text{CH}_2$	heptane	85/71	<10/<10
Comparative Example 6	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	heptane	103/88	<10/<10
Comparative Example 7	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	none	110/101	25/24
Comparative Example 8	$\text{CH}_3(\text{CH}_2)_{17}\text{CH}=\text{CH}_2$	heptane	106/96	<20/<20
Comparative Example 9	$\text{CH}_3(\text{CH}_2)_{17}\text{CH}=\text{CH}_2$	none	111/92	41/38
Comparative Example 10	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	none	112/99	35/30
Comparative Example 11	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	heptane	105/91	<10/<10
Comparative Example 12	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	none	109/98	<20/<20
Comparative Example 13	$\text{CH}_3(\text{CH}_2)_{13}\text{CH}=\text{CH}_2$	heptane	105/91	<10/<10

Example	Olefin	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
13	E7	heptane	96/81	31/24
14	E9	heptane	87/69	30/20
15	E10	heptane	101/82	48/40
16	E1	heptane	112/107	57/52
17	E1	SHFE	105/92	57/51
18	E1	SHFE	97/78	47/39
19	E1	heptane	126/93	66/34
20	E1	heptane	100/66	<20/<20
21	E1	heptane	100/79	38/30

### EXAMPLE 22

A silicon chip, prepared as in Example 12, containing MEMS cantilever beams and having the photoresist removed, was placed in the "fill / drain" apparatus. A volume of 5 mL CMOS grade 49 percent by weight aqueous HF was added to the apparatus and allowed to etch the chip for 3 min to release the cantilever beam MEMS devices. The chip was then washed with 10 x 25 mL portions of deionized water ( $\geq 18$  megohms resistance) over a total period of 10 min. After the last of the water was drained, 5 mL of CMOS grade 49 percent by weight aqueous HF was added and allowed to etch the MEMS chip again for 3 min. The aqueous HF was again drained. The etched chip was then sequentially washed in isopropyl alcohol for 10 min, and heptane for 10 min, each wash being performed in the same manner as the water wash, with 10 fill and drain cycles over 10 min.

The chip was removed from the heptane, and placed in a quartz tube in a solution (sparged with dry nitrogen for 15 min) containing 5 parts by weight of E1, 5 parts by weight 2-hydroxy-2-methyl-1-phenyl-1-propanone (having the trade designation "DAROCUR 1173") free-radical photoinitiator, and 90 parts by weight heptane. The tube was fitted with a rubber septum and irradiated for 30 min under the conditions of Example 11. After irradiation, the MEMS chip was removed from the quartz tube and placed in a small glass cup (approximately 2 mL volume) containing heptane. The MEMS chip was washed by adding 30 mL of heptane over a period of 10 min to the cup (excess overflowed). This was followed by addition of 30 mL isopropyl alcohol to the cup over

10 min time, followed by addition of 30 mL SHFE to the cup over 10 min time. The cup containing the MEMS chip (under SHFE) was placed in an oven at 140 °C for 10 min, thereby drying the MEMS chip. According to this procedure, cantilever beams having a length of 1150 micrometers long were observed (according to the method of Example 12) to be free from the surface, with longer beams showing evidence of stiction.

### EXAMPLE 23

A chip as in Example 12, containing MEMS cantilever beams and having the photoresist removed, was immersed in CMOS grade 49 percent by weight aqueous HF for 3 min to release the cantilever beam MEMS devices. The chip was then washed with deionized water ( $\geq 18$  megohms resistance) for 10 min. The etched chip was then washed in isopropyl alcohol for 2 min, followed by another isopropyl alcohol rinse for 3 min. The chip was transferred to a third isopropyl alcohol bath in a small dish (~2 mL volume) used for coating the sample. The isopropyl alcohol was displaced from the dish containing the chip with 10 x 2 mL portions SHFE. The SHFE was then displaced with 6 x 2 mL portions of a 1.0 millimolar solution of E10 in SHFE (approx. 90 g) containing two drops of a solution of 1 g of chloroplatinic acid in 10 mL of isopropyl alcohol and held for 1 min. The coating solution was then displaced with 10 x 2 mL portions of SHFE. The MEMS chip was dried by placing the sample dish containing the chip under SHFE into a heated 10 mL high-pressure stainless steel vessel equipped with a high-pressure pump at the inlet and a vent line at the outlet. The vessel was filled with CO<sub>2</sub> at 60 °C and 200 atmospheres pressure (20 megaPascals)). The contents of the vessel were displaced with 5 x 10 mL portions of CO<sub>2</sub>, effectively removing the SHFE. The CO<sub>2</sub> remaining in the vessel was vented to atmospheric pressure and the dry chip was removed for evaluation according to the method of Example 12. Cantilever beams having a length of 1450 micrometers were observed to be free from the surface, with longer beams showing evidence of stiction.

### EXAMPLES 24-25 and COMPARATIVE EXAMPLE 14

Examples 24-25 and Comparative Example 14 were carried out according to General Procedure D. The specific fluorinated olefin used, solvent, and results of contact angle measurements are reported in Table 3 (below)

TABLE 3

Example	Olefin	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
Comparative Example 14	$\text{CF}_3(\text{CF}_2)_7\text{CH}=\text{CH}_2$	FSOLV	109/98	58/50
24	E1	FSOLV	111/84	59/52
25	E8	FSOLV	119/106	63/58

5

EXAMPLES 26-30

Effects of hydrosilylation process conditions for fluorinated olefin E9 are reported in Examples 26-30. General Procedure A was used to prepare the silicon substrate used in Examples 26-30.

10 Example 26 was carried out according to General Procedure B, except that neat E9 was placed in a test tube having a ground glass joint and fitted with a condenser. Prior to introduction of the silicon wafer piece, the E9 in the test tube was sparged for 30 min with nitrogen. After the sample was introduced, nitrogen sparging was continued for 15 min.

15 Example 27 was carried out according to General Procedure D with the following changes: neat E9 was used (that is, no solvent); prior to introduction of the silicon wafer piece, the E9 in the test tube was sparged for 1 hr with nitrogen; after the silicon wafer piece was introduced to the tube, nitrogen sparging was continued for 2.5 hr, then the sample was heated in an oil bath at 180 °C for 30 min.

20 Example 28 was carried as in Example 27 except that a 5 percent by volume solution of E9 in FSOLV was used in place of neat E9.

Example 29 was carried out as in Example 26, except that no condenser was utilized.

25 Example 30 was carried out using a glass apparatus consisting of three pieces of tubing joined in the shape of an "H". The top of each part of the "H" was open for sample and reagent introduction. The "crossbar" of the "H" was 2 cm above the bottoms of the "H". Into one leg of the "H" was placed the Si wafer piece, and into the other was placed 1 mL of E9. The etched silicon wafer piece having Si-H bonds was placed in the dry side

of the "H" after the neat compound in the wet side had been sparged for 30 min with nitrogen. The tops of the "H" were sealed with rubber septa, and the apparatus was purged for an additional hour with nitrogen. The outlet tube for the nitrogen purge was removed, but the inlet tube was left on so the apparatus remained under a positive pressure of nitrogen. The apparatus was then placed in the heat bath according to General Procedure B for 4.5 hr, removed and cooled for 5 min using room temperature water. The wafer piece was removed and sequentially washed with SHFE, heptane, acetone, isopropyl alcohol, and blown dry.

Results of contact angle measurements, reported in Table 4 (below), can be compared to results reported, for example, in Comparative Example 1 and Example 9.

TABLE 4

Example	Process Conditions	Solvent	Contact Angle advancing/receding, (degrees)	
			Water	Hexadecane
26	200 °C, 3 hr	neat	59/<10	<10/<10
27	180 °C oil bath, 30 min	neat	91/72	39/29
28	180 °C oil bath, 30 min	FSOLV	80/61	33/23
29	200 °C, 3 hr	neat	68/29	<10/<10
30	200 °C, 4.5 hr	none, vapor phase	93 / 64	46 / 37

## EXAMPLES 31-34

Examples 31-34 illustrate the effect of hydrosilylation process conditions for fluorinated olefins E12 and E13.

Example 31 was carried according to General Procedure D, except that neat E12 was used (no solvent), and the tube was heated at 180° C for 30 min.

Example 32 was carried out as in Example 30, except that 100 milligrams of E12 was used as the olefin instead of E9.

Example 33 was carried out as in Example 31, excepting that E13 was used instead of E12.

Example 34 was carried out as in Example 30, except that E13 was used instead of E9.

Results of contact angle measurements are reported in Table 5 (below).

TABLE 5

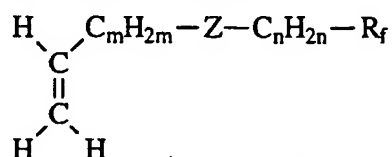
Example	Olefin	Process Conditions	Phase	Contact Angle advancing/receding, (degrees)	
				Water	Hexadecane
31	E12	180 °C, 30 min	liquid	84/39	43/31
32	E12	200 °C, 4.5 hr	vapor	106/76	54/47
33	E13	180 °C, 30 min	liquid	83/43	41/28
34	E13	200 °C, 4.5 hr	vapor	100/64	51/45

5

What is claimed is:

1. A method for modifying a silicon substrate comprising:  
providing a silicon substrate having a plurality of Si-H bonds;  
5 providing a composition comprising:

a fluorinated olefin having the formula:



wherein

m is an integer greater than or equal to 1,

n is an integer greater than or equal to 0,

Z is a divalent linking group,

R<sub>f</sub> is a highly fluorinated organic group, and

optional solvent; and

contacting the composition with the silicon substrate under conditions such that the  
15 fluorinated olefin becomes covalently attached to the surface of the silicon substrate.

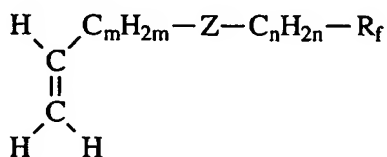
2. An article comprising a chemically modified silicon substrate prepared according to a method comprising:

providing a silicon substrate;

20 etching the silicon substrate to form a plurality of Si-H bonds;

providing a composition comprising:

a fluorinated olefin having the formula:



wherein

m is an integer greater than or equal to 1,

n is an integer greater than or equal to 0,

Z is a divalent linking group,

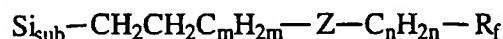
$R_f$  is a highly fluorinated organic group, and

optional solvent; and

contacting the composition with the silicon substrate under conditions such that the fluorinated olefin becomes covalently attached to the silicon substrate.

5

3. An article comprising a chemically modified silicon substrate, wherein the chemically modified substrate comprises at least one silicon atom that is covalently bonded to an organic group having the formula:



10

wherein

$m$  is an integer greater than or equal to 1;

$n$  is an integer greater than or equal to 0;

$Z$  is a divalent linking group;

$R_f$  is a highly fluorinated organic group; and

15

$Si_{sub}$  is a substrate silicon atom directly bonded to at least one additional substrate silicon atom.

4. The method of claim 1 or the article of claim 2 or claim 3, wherein  $m$  is in the range of from 1 to 20.

20

5. The method of claim 1 or the article of claim 2 or claim 3, wherein  $m$  is 1.

6. The method of claim 1 or the article of claim 2 or claim 3, wherein  $n$  is in the range of from 0 to 20.

25

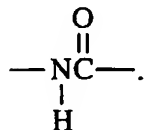
7. The method of claim 1 or the article of claim 2 or claim 3, wherein  $n$  is in the range of from 0 to 3.



8. The method of claim 1 or the article of claim 2 or claim 3, wherein Z is a covalent

bond,  $-O-$ , or  $\begin{array}{c} O \\ || \\ -NC- \\ | \\ R^1 \end{array}$ , wherein  $R^1$  is alkyl or H.

9. The method of claim 1 or the article of claim 2 or claim 3, wherein Z is  $-O-$  or



5

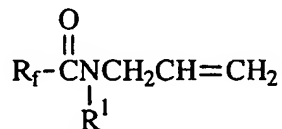
10. The method of claim 1 or the article of claim 2 or claim 3, wherein  $R_f$  contains fluorine in an amount of greater than or equal to at least 50 percent by weight.

11. The method of claim 1 or the article of claim 2 or claim 3, wherein  $R_f$  contains  
10 fluorine in an amount of greater than or equal to at least 60 percent by weight.

12. The method of claim 1 or the article of claim 2 or claim 3, wherein  $R_f$  comprises a perfluoroalkyl group having from 4 to 12 carbon atoms.

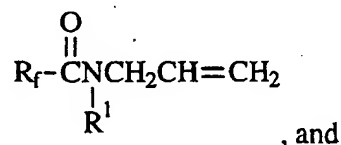
13. The method of claim 1 or the article of claim 2, wherein the fluorinated olefin  
15 comprises a poly(perfluoroalkyleneoxy) group.

14. The method of claim 1 or the article of claim 2, wherein the fluorinated olefin has the formula:



20

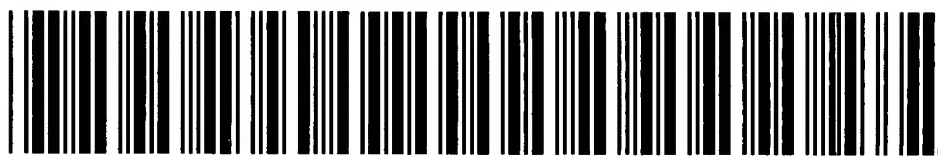
15. The method of claim 1 or the article of claim 2, wherein the fluorinated olefin has the formula:



wherein the fluorinated olefin comprises a poly(perfluoroalkylenoxy) group.

16. The method of claim 1 or the article of claim 2, wherein the fluorinated olefin  
5 comprises a plurality of allyloxy groups.
17. The method of claim 1 or the article of claim 2 or claim 3, wherein the silicon  
substrate comprises a microelectromechanical systems device.
- 10 18. The method of claim 1 or the article of claim 2 or claim 3, wherein the silicon  
substrate comprises a microfluidic device.
19. The method of claim 1 or the article of claim 2, wherein the composition further  
comprises solvent having a boiling point of greater than or equal to 180 °C.  
15
20. The method of claim 1 or the article of claim 2, wherein the conditions comprise a  
temperature in excess of 180 °C.
21. The method of claim 1 or the article of claim 2, wherein contacting the  
20 composition with the silicon substrate comprises contacting a fluorinated olefin vapor with  
the silicon substrate.
22. The method of claim 1 or the article of claim 2, wherein the composition further  
comprises at least one of a free-radical photoinitiator or a photocatalyst; and wherein said  
25 conditions comprise electromagnetic radiation.
23. The method of claim 22, wherein the free-radical photoinitiator comprises 2-  
hydroxy-2-methyl-1-phenyl-1-propanone.

# IDS REFERENCES



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(19) World Intellectual Property Organization  
International Bureau



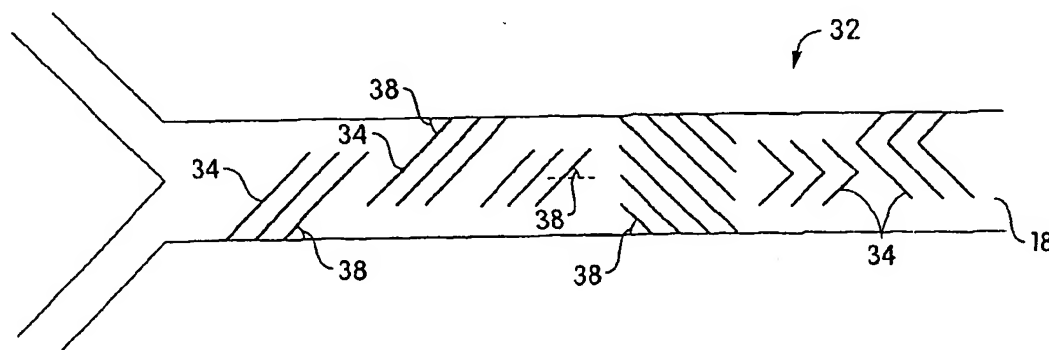
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(54) Title: LAMINAR MIXING APPARATUS AND METHODS



(57) Abstract: A mixing apparatus is used to effect mixing between one or more fluid streams. The mixing apparatus generally functions by creating a transverse flow component in the fluid flowing within a channel without the use of moving mixing elements. The transverse or helical flow component of the flowing fluid or fluids can be created by weak modulations of the shape of the walls of the channel. Transverse or helical flow component can be created by grooves features defined on the channel wall. Specifically, the present invention can be used in laminarly flowing fluids. The mixing apparatus and methods thereof can effect mixing of a fluid or fluids flowing with a Reynolds number of less than about 100. Thus, the present invention can be used to mix a fluid flowing in the micro-regime. The mixing apparatus can be used to mix a fluid in a microfluidic system to significantly reduce the Taylor dispersion along the principal direction. The mixing apparatus can be used to increase the effective exposed area to promote diffusion of components between or within the fluid or fluids.

WO 03/011443 A2

## LAMINAR MIXING APPARATUS AND METHODS

### Background of the Invention

#### 1. Field of the Invention

5 The present invention relates to mixing laminarly flowing fluids and, more particularly, to low Reynolds number mixing apparatus and to methods of use thereof.

#### 2. Description of Related Art

Mixers are known in the art for mixing materials. These mixers may be useful  
10 in various applications such as mixing chemicals in industrial processes, mixing multi-part curing systems in adhesives, foams and molding compounds, mixing fuels and gases for combustion, mixing air into water for sewerage treatment, or wherever mixing needs to be accomplished.

There are generally two types of fluid flow, laminar flow and turbulent flow. In  
15 laminar flow, the fluid flows in smooth layers or lamina. This occurs when adjacent fluid layers slide smoothly over one another with mixing between layers or lamina occurring predominantly on a molecular level by diffusion. Turbulent flow is characterized by fluctuations of the velocity of the fluid in both space and time. Mixing of two or more substances in turbulent flow conditions generally proceeds faster than  
20 under laminar flow conditions.

The viscosity, the flow rate, and the density of the fluid along with the diameter of the flow path dictates the type of fluid flow. The more viscous two materials are or the smaller the cross-sectional dimension of the channel in which they flow, the higher the flow rate required in order to create a turbulent flow. These variables can be  
25 combined into a dimensionless parameter to characterize the flow called the Reynolds number according to

$$\text{Re} = \frac{D\rho v}{\mu}$$

where D is the characteristic dimension of the path,  $\rho$  is the density of the fluid,  $v$  is the  
30 fluid flow velocity, and  $\mu$  is the viscosity of the fluid. Flows are typically laminar for Re less than 2300 and turbulent for Re less than 2300.

### Summary of the Invention

In one embodiment, the present invention relates to an article. The article comprises a microfluidic channel defined therein and designed to have fluid flow therethrough in a principal direction. The microfluidic channel includes a channel surface having at least one groove or protrusion defined therein. The at least one groove or protrusion has a first orientation that forms an angle relative to the principal direction.

In another embodiment, the present invention provides an article comprising a microfluidic channel constructed and arranged to have a fluid flowing therethrough while creating a transverse flow component in the fluid.

In another embodiment, the present invention relates to an article comprising a structure having a channel defined therein, the channel designed to have a fluid flowing therethrough in a principal direction, the channel including a channel surface having a plurality of chevron-shaped grooves or protrusions formed in at least a portion of the channel surface so that each chevron-shaped groove or protrusion has an apex that defines an angle.

In yet another embodiment, the present invention relates to a structure. The structure comprises a first channel having a width that is less than about 1000  $\mu\text{m}$ , a second channel having a width that is less than about 1000  $\mu\text{m}$  and a third channel having a principal direction and a width that is less than about 1000  $\mu\text{m}$ . The third channel connects the first and second channels and comprises channel surfaces having grooves or protrusions defined therein. The grooves or protrusions are oriented at an angle relative to the principal direction.

In another embodiment, the present invention relates to a method for dispersing a material in a fluid. The method comprises the steps of providing an article having a channel designed to have fluid flow therethrough in a principal direction, the channel including a channel surface having at least one groove or protrusion therein that traverses at least a portion of the channel surface, at least one groove or protrusion oriented at an angle relative to the principal direction and causing the fluid in the channel to flow laminarily along the principal direction.

In another embodiment, the present invention is directed to a method. The method comprises the steps of causing a first fluid to flow in a channel at a Reynolds number that is less than about 100, causing a second fluid to flow in the channel at a

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Reynolds number that is less than about 100 and creating a transverse flow component in the first and the second fluids to promote mixing between the first and second fluids.

In yet another embodiment, the present invention is directed to a method for forming a microfluidic article. The method comprises the steps comprising forming a first topological feature that has a smallest dimension that is less than about 1000  $\mu\text{m}$  on a surface of a mold substrate, forming a second topological feature on the first topological feature to form a mold master, the second topological feature characterized by a length that traverses at least a portion of a section of the first topological feature, placing a hardenable material on the surface, hardening the material thereby creating a molded article having a microfluidic channel shaped from the first topological feature and at least one groove or protrusion shaped from the second topological feature and removing the microfluidic article from the mold master.

In another embodiment, the present invention is directed to a method for producing a helical flow path in a fluid flowing along a principal direction. The method comprises the step of providing a structure having a surface with a plurality of substantially linear grooves or protrusions oriented at an angle relative to the principal direction. The grooves or protrusions are formed to be parallel to and periodically spaced from each other. The method further comprises the step of causing the fluid to flow along the surface. The fluid flowing adjacent the surface has a Reynolds number that is less than about 100.

Other advantages, novel features, and objects of the invention will become apparent from the following detailed description of the invention when considered in conjunction with the accompanying drawings, which are schematic and which are not intended to be drawn to scale. In the figures, each identical, or substantially similar component that is illustrated in various figures is represented by a single numeral or notation. For purposes of clarity, not every component is labeled in every figure, nor is every component of each embodiment of the invention shown where illustration is not necessary to allow those of ordinary skill in the art to understand the invention.

30

#### **Brief Description of the Drawings**

FIG. 1 is a schematic diagram of one embodiment of the present invention

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illustrating a system with channels defined in a substrate;

FIG. 2a is a schematic diagram showing a perspective view of one embodiment of the mixing apparatus with a fluid flowing therethrough;

FIG. 2b is an elevational view of the embodiment of FIG. 2a illustrating the  
5 grooves defined on a channel wall thereon;

FIG. 3 is a schematic diagram of one embodiment of the invention showing a channel having various configurations of grooves;

FIG. 4a is a schematic diagram of one embodiment of the invention showing a top elevational view of a mixing apparatus having grooves;

10 FIG. 4b is a diagram of the apparatus of FIG. 4a along b-b schematically showing the transverse or helical flow component of a flowing fluid;

FIG. 4c is a copy of a micrograph showing the transverse or helical flow component created within a fluid flowing in a mixing apparatus having grooves according to one embodiment of the present invention;

15 FIG. 5 is a schematic diagram of one embodiment of the present invention illustrating a mixing apparatus having chevron-shaped grooves defined on a wall therein;

FIGS. 6a-6f are copies of micrographs illustrating the cross-section of the mixing apparatus of FIG. 5 having two fluids flowing therethrough at different points  
20 along the length of the mixing apparatus;

FIG. 7 is a graph showing how the number of cycles affects the standard deviation of intensity, as a measure of mixing progress;

FIG. 8 is schematic diagram showing the dispersion of a plug of miscible solution along the principal direction of flow without (top) and with (bottom)  
25 continuous mixing according to one embodiment of the present invention; and

FIG. 9 a-b are copies of micrographs showing the difference between axial dispersion without (FIG. 9a) and with (FIG. 9b) mixing according to one embodiment of the present invention.

30 **Detailed Description**

The present invention is directed to mixing apparatus and methods used to effect mixing between one or more fluid streams. The mixing apparatus generally



- 5 -

functions by creating a transverse flow component in the fluid flowing within a channel without the use of moving mixing elements. The transverse or helical flow component of the flowing fluid can be created by the shape of the channel walls. For example, the transverse component can be created by grooves defined on the channel wall. The present invention can be used in systems where diffusion primarily controls fluid mixing. The term "transverse" is meant to describe a crosswise direction or at angle relative to a direction of a channel and the term "helical" is meant to describe a continuous plane curve that is extended in one direction and periodic in the other two. The term "principal direction" is meant as the direction of flow along a flow structure through which the bulk or the majority of the fluid can flow. For example, in a channel, the principal direction typically along the length of the channel, in contrast to across the width of the channel. Thus, the term "transverse flow component" is meant to describe a flow component that is oriented at an angle relative to a particular direction, preferably, relative to the principal direction. Notably, the present invention can be particularly useful when used in connection with microfluidic systems.

Patterned topography on surfaces according to the present invention can be used to generate chaotic flows in contexts other than pressure driven flows in microchannels. For example, chevron-shaped structures on the walls of round pipes and capillaries can provide efficient mixing. Thus, in one embodiment, fluid unit operation dependent on heat or mass transfer, such as a heat exchanger, may have turbulent flow in the bulk flowing fluid but may incorporate grooves, in a variety of geometries, on baffle plates to reduce or at least partially eliminate boundary limiting conditions that typically affect the overall transfer coefficient. That is, chaotic flows will also exist in the laminar shear flow in the boundary layer of an extended flow over a surface that presents the staggered herringbone features. This stirring of the boundary layer will enhance the rates of diffusion limited reactions at surfaces (e.g. electrode reactions) and heat transfer from solids into bulk flows. In another embodiment, electroosmotic flows in capillaries that contain the staggered herringbone features can be chaotic and promote stream mixing.

FIG. 1 illustrates a microfluidic system 10 according to one embodiment of the present invention. System 10 includes a substrate 12 with a surface 14 having formed or defined therein a structure 16 that can be a part of a network or array (not shown) of

- 6 -

similar and interconnected structures and features. Structure 16 includes a channel 18 formed on surface 14 of substrate 12, a source 20 at a first position 22 that can provide a fluid 24 flowing in channel 18 and a sink 26 at a second position 28 wherein fluid 24 is received.

5           In another aspect, the present invention functions, in part, by increasing the effective exposed or interfacial area to promote diffusion of components between distinct volumes of the flowing fluid. That is, the present invention, in one embodiment, promotes mixing by diffusion by diverting a portion of the flowing fluid as, for example, by creating a transverse flow component in the flowing fluid. The  
10   transverse flow component may create a "folding effect" so that the effective exposed area through which diffusion of molecular species can occur is increased or, in another sense, the distance over which diffusion must act to eliminate concentration variations is decreased. Such an effect may reduce the rate of dispersion along the flow by carrying unit volumes of the fluid between fast and slow moving regions. In net effect,  
15   i.e., as the fluid progresses through the mixing apparatus, the mixing of the fluid or fluids is increased as the diffusion area is increased and, consequently, the time required to achieve mixing to a desired homogeneity is reduced. The transverse flow component may be viewed, analogously, to the effect created by turbulent flow wherein localized eddy currents are created as a consequence thereof. In another aspect, the  
20   transverse flow component can be viewed as stretching the volumes of the fluid at an exponential rate as the fluid is "wound" helically along the principal direction of the flow.

          The present invention can be used in laminarly flowing fluids. Thus, as described below, the mixing apparatus and methods thereof are particularly suitable to  
25   mix a fluid flowing in the micro-regime. As used herein, the term "microchannel" refers to a channel that has a characteristic dimension, i.e., a width or a depth, that is less than about 1000 microns ( $\mu\text{m}$ ). System 10 can be used to mix a fluid or fluids in a microfluidic system to significantly reduce the Taylor dispersion along the principal direction. The present invention may be used advantageously in microfluidic systems  
30   wherein the laminar flow is particularly predominant. Fluids flowing in such systems are typically characterized as laminar Poiseuille flows with low Reynolds numbers. As described further below, the mixing apparatus can be designed to create a transverse

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flow component within such flows that are non-turbulent, preferably with Re having a Reynolds number that is less than about 2000, preferably, less than 100, more preferably, less than about 12, and even more preferably, less than 5.

Thus, in one embodiment, grooves or protrusions can be oriented in a variety of configurations or combinations to effect transverse flow components of the fluid or fluids flowing therethrough that is independent of Reynolds number or as Reynolds number goes to zero.

The present invention, as embodied in the schematic illustration of FIG. 1, can be used in a system wherein a desired process operation may be carried out including, but not limited to, flowing a fluid, facilitating a chemical reaction, dissolving a substance in a medium, depositing or precipitating a material on a surface, mixing a fluid or fluids to achieve homogeneity and exposing a first material to a second material. For purposes of illustration, a system 10, as shown in FIG. 1, will be described with respect to a flowing fluid. As used herein, fluid can refer to a gas or a liquid.

According to one embodiment, channel 18 can be formed as a mixing apparatus 32 to facilitate mixing a fluid or fluids flowing therethrough. As schematically illustrated in the embodiment of FIG. 2a, channel 18 comprises a mixing apparatus 32 having a rectangular cross-section with a width and a depth or height. Grooves, undulation or protrusion features 34 are formed on at least one channel surface 30. Fluid 24 flowing in channel 18 has a principal direction, indicated by reference 36, along the lengthwise direction of the channel. In other embodiments, the microfluidic channel can have a variety of cross-sectional shapes including, but not limited to, rectangular, circular and elliptical.

In some embodiments, the groove is oriented to form an angle relative to the principal direction. Grooves 34 on channel surface 30 are constructed and arranged to create an anisotropic response to an applied pressure gradient thereby producing at least one three-dimensional flowpath such as transverse flow component in fluid 24 flowing in channel 18. Grooves 34 can be formed as undulations that provide reduced flowing resistance along the valleys 40 of grooves 34. That is, fluid near channel surface 30 having groove 34 is exposed to reduced flow resistance at or near the valleys 40 creating a transverse flow component 42. As the fluid flows further along principal

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direction 36, transverse flow components 42 are further generated or increase in magnitude through additional grooves 32 defined along channel surface 30. The resultant effect creates a circulating or helical flow path 44.

Grooves 34 typically have a width and a height that is less than the width and height of mixing apparatus 32 and can be arranged periodically along the lengthwise direction of mixing apparatus 32. As shown in the schematic illustration of FIG. 3, grooves 34, defined on channel surface 30 of mixing apparatus 32, can have a variety of configurations and combinations. That is, in one embodiment, grooves 34 can be oriented at an angle 38 and can extend substantially or partially across the cross-section of mixing apparatus 32. Further, it can be seen that those of ordinary skill may recognize that grooves 34 can have a variety of geometrical cross-sections including, but not limited to rectangular, circular and parabolic. Grooves or protrusions 34 can be oriented in a variety of configurations or combinations to effect transverse flow components of the fluid or fluids flowing therethrough that is independent of Reynolds number or as Reynolds number goes to zero.

In another embodiment, grooves 34 can be arranged as a set of grooves, wherein each groove is arranged periodically as shown in FIGS. 3-5. Thus, in one embodiment, the mixing apparatus can comprise at least one set, preferably at least two sets and more preferably, a plurality of sets wherein each set comprises a plurality of grooves arranged periodically therein. In another embodiment, each set comprises a periodic arrangement of grooves that are offset from each other such that at least one set is at least partially coextensive with at least another set. In another embodiment, the mixing apparatus comprises a set comprising a plurality of grooves having various configurations. Thus, as illustrated in FIG. 3, the grooves may be oriented at an angle relative to the principal direction, may be offset, traverse at least a portion of the cross-section of the mixing apparatus, may be periodically arranged to form a set or a repeating cycle and may have chevron shapes. Chevron-shaped structures typically have at least one apex, which is formed by lines intersecting at an angle. The term "chevron-shape" is meant to represent a structure having a V-shape or zigzag shape. And, as used herein, the term "chevron-shaped" is meant to include structures formed by intersecting linear and non-linear lines as well as symmetrical and asymmetrical V-shapes and structures having multiple intersections.

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In one embodiment, the mixing apparatus comprises herringbone-shaped or chevron-shaped features that are asymmetric with respect to a lengthwise axis of the channel forming the mixing apparatus. In another embodiment, the asymmetry of the chevron-shaped features vary in alternating or in other predetermined fashion. For  
5 example, with reference to FIG. 5, the asymmetry of chevron-shaped grooves in the first set differs from that of the adjacent set.

As used, herein, a pair of sets forms a cycle of the mixing apparatus. The term "cycle" is refers to a plurality of sets that are sufficient to produce a spiral flow component. Thus, in one embodiment, one cycle refers to a first set of similarly  
10 grooves and a second set of similarly shaped grooves. A set of cycles may comprise a plurality of cycles, each cycle comprising sets of shaped features and each cycle may be geometrically distinguishable from another cycle. For example, a set may comprise a group of chevron-shaped grooves defining a first apex group that are similarly shaped and a second set of chevron-shaped grooves defining a second apex group that are  
15 similarly shaped, the second apex group are "offset" from the first apex group such that the apex is displaced from the first group relative to an axis, e.g., the axis along the principal direction. Such a design can be characterized by, among others, the degree of asymmetry as measured by the fraction of the width of the channel that is spanned by the wider branch of the chevron-shaped grooves and the amplitude of the rotation of the  
20 fluid, as measured by  $\theta$  and shown in FIG. 4b, that is induced by the chevron-shaped structures. The amplitude of the rotation is influenced by the geometry of the undulations and the number of undulations per set or half cycle.

Thus, in another embodiment, the mixing apparatus comprises a first channel disposed in a structure having a width that is less than about 5000  $\mu\text{m}$ , a second channel  
25 also disposed in the structure and also having a width that is less than about 5000  $\mu\text{m}$  and a third channel with a principal direction and having a width that is less than about 5000  $\mu\text{m}$  that connects the first and second channels and comprising channel surfaces with grooves, which are oriented at an angle relative to the principal direction. However, those of ordinary skill practicing the invention may readily recognize that the  
30 structures described herein may be used to effect mixing in any non-turbulent flow system. Thus, a system that may have a relatively large characteristic dimension may nonetheless be non-turbulent if the fluid flowing therein or the fluid flowing adjacent to

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the features that create a transverse flow component are non-turbulent. For example, mixing may be effected by creating a transverse flow component, with a use of grooves, in a fluid flowing on a surface that extends essentially infinitely in two dimensions. Notably, the fluid may be flowing non-turbulently adjacent to the grooves but may be flowing turbulently away from the surface. Thus, the invention may be used in a surface or a mixing apparatus regardless of the dimension of the channel.

The staggered herringbone mixing apparatus based on patterned topography on the surface of microchannels can offer a general solution to the problem of mixing fluids in microfluidic systems. The simplicity of its design allows it to be easily integrated into microfluidic structures with standard microfabrication techniques. Such a mixing apparatus can operate over a wide range of  $Re$ , specifically, all values less than about 100.

Substrate 12 can be formed from any suitable material that can be used to create structures 16 and performing the desired process operation. Substrate 12 can be formed of a polymeric material such as a random or block polymeric or copolymeric material; suitable polymeric materials include polyurethane, polyamide, polycarbonate, polyacetylene, polysiloxane, polymethylmethacrylate, polyester, polyether, polyethylene terephthalate and/or blends or combinations thereof. Substrate 12 can also be a ferrous, non-ferrous, transition or precious metal such as steel, platinum, gold and/or alloys or combinations thereof. Substrate 12 can be formed of a semiconductor material such as silicon and gallium arsenide including nitrides and oxides formed thereof. The selection of materials suitable to create structures and perform the desired process operation can be performed by those of ordinary skill practicing the field.

Systems of the present invention can be prepared using soft lithographic techniques. One such technique is discussed by McDonald et al. in *Electrophoresis* 21, 27-40 (2000), which is incorporated in its entirety. Master structures are typically made with two step photolithography, which generally involves preparing a first photolithographic layer defining a positive image of the channel or mixing apparatus and a second photolithographic layer defining a positive image of the pattern of grooves or undulation. The first photolithographic layer can be used as a positive image of the channel. The second layer can be used as a positive image of the pattern of undulations. This second pattern is typically aligned to lie on top of the channel

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using a mask aligner. The master structures can then be used as molds to create a substrate made from polydimethylsiloxane (PDMS).

To close the molded channel, the PDMS substrates are typically exposed to plasma for one minute and can then be sealed with a glass cover slip. The thickness of the cover slip is typically selected to be optically compatible with the oil immersion objectives of the confocal microscope. For example, a No. 1 glass cover slip can be used with a XX Leica confocal microscope with a 40x/1.0n.a. objective. It should be understood that other techniques can also be used to form systems of the present invention.

The functions and advantages of these and other embodiments of the present invention can be further understood from the examples below. The following examples are intended to illustrate the benefits of the present invention, but do not exemplify the full scope of the invention.

#### Example 1

This example, with reference to FIGS. 4a-c, discusses one embodiment of the present invention and is directed to mixing fluids in a mixing apparatus. The broad dark lines, shown in FIG. 4a, represent undulations in the channel surface. A sequential pair of grooves form one cycle. The grooves were oriented at a 45-degree angle relative to the principal direction. Mixing apparatus 32 was a microfluidic article with a rectangular cross-section, which was about 200  $\mu\text{m}$  wide and comprised a plurality of fluid inlets 46, 48 and 50, a plurality of sets 52 of grooves comprised a cycle, each set with at least one groove 34 arranged periodically along the principal direction. Fluids 54, 56, and 58 were introduced through inlets 46, 48 and 50, respectively wherein fluid 56 is comprised a fluorescent dye. As the fluids flowed laminarly at a Reynolds number that is less than about 100, a transverse flow component 42 was created in the aggregated fluid in mixing apparatus 32 as schematically depicted in FIG. 4b and as shown in the copy of a micrograph in FIG. 4c. The lighter portions represent the fluorescent dye introduced in fluid 56. This example showed that the grooves in the mixing apparatus can create transverse flow components in a fluid having a Reynolds number that is less than about 100.

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Example 2

This example, with reference to FIGS. 5 and 6a-c, discusses another embodiment of the present invention and directed to mixing fluids in a mixing apparatus, specifically, a staggered herringbone mixing apparatus having chevron-shaped grooves. The broad dark lines, shown in FIG. 5, represent the chevron-shaped undulations in the channel surface. A sequential pair of grooves form one cycle. The grooves were oriented at a 45-degree angle relative to the principal direction. The mixing apparatus was a microfluidic article with a rectangular cross-section, which was about 200  $\mu\text{m}$  wide by 100  $\mu\text{m}$  tall and comprised a plurality of fluid inlets 48 and 50, a plurality of sets 52 of 50  $\mu\text{m}$  x 50  $\mu\text{m}$  rectangular chevron-shaped grooves comprised a cycle, each set with six chevron-shaped groove 34 arranged periodically along the principal direction. The sets were disposed from each other such that the loci of apex of one set was offset from the loci of apex of an adjacent set. Fluids 56 and 58 were introduced through inlets 48 and 50 respectively from their respective reservoirs (not shown). Fluid 56 comprised a fluid, poly(ethylenimine), MW 750,000, fluorescently labeled with 1% FITC in 0.1 wt. % solution while fluid 58 comprised the same solution without FITC. The fluids were pumped through the mixing apparatus at a velocity of about 2.7 cm/s by applying a constant pressure on each fluid reservoir with compressed air. The corresponding Reynolds number was determined to be about  $4 \times 10^{-2}$  and the Péclet number was determined to be about  $3.3 \times 10^{+4}$ .

FIGS. 6a-f are copies of micrographs of vertical cross-sections along the mixing apparatus made using a XX Leica confocal microscope with a 40x/1.0n.a. objective. These show the distribution of the fluorescent molecules before the first cycle (FIG. 6a), and progressively after the first (FIG. 6b), second (FIG. 6c), fourth (FIG. 6d), eight (FIG. 6e) and sixteenth cycles (FIG. 6f). FIGS. 6b-f shows that generation of a transverse flow components (depicted by the lighter portions) in the fluid as the fluid flows through multiple cycles. Notably, the fluid appears homogeneous after the sixteenth cycle. Thus, this example shows that a mixing apparatus having chevron-shaped grooves can be used to mix fluids flowing at very low Reynolds numbers.



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Example 3

In this example, the efficiency of mixing was evaluated. Four fluids were prepared with fluorescent pigment similar to the fluids described in Example 2. The fluids were introduced under varying conditions into a mixing apparatus. The fluids  
5 flowed with a Reynolds number that was less than about 7.5 and, respectively, with Péclet numbers of  $1.6 \times 10^2$  (circle),  $1.9 \times 10^2$  (square),  $7.4 \times 10^3$  (triangle) and  $3.3 \times 10^4$  (diamond). Péclet number is the product of the Reynolds and Prandtl numbers. The latter is the viscosity,  $\mu$ , of a fluid divided by its molecular diffusivity. Thus, the Péclet number is

$$10 \quad Pe = \frac{Uh}{D}$$

where  $U$  is the average velocity,  $h$  is the height of the channel and  $D$  is the diffusivity of the diffusing material in the medium.

Fluorescence intensity was found to be proportional to the concentration of fluorescent molecules and accordingly, mixing efficiency was characterized as the  
15 variation of intensity of the fluorescence. Stated another way, as the degree of mixing increases, the variation measured as the standard deviation of fluorescent intensity approaches zero. FIG. 7 is a chart showing the standard of deviation of intensity relative to the number of cycles for fluids having various Péclet numbers. As expected, a fluid with a lower Péclet number required less mixing cycles than a fluid with a  
20 higher Péclet number because diffusion was the predominant mechanism of mixing. The standard deviation approached 20, not zero, because, it is believed, of optical effects, shadows in the field of view of the microscope, and the noise of the photodetector.

The example shows that the number of mixing cycles that are required for total  
25 mixing grows slowly with Péclet number but that the mixing apparatus according to the present invention can be used to efficiently mix laminarly flowing fluids. The inset shows that the number of cycles required for total mixing is linearly proportional to  $\log(Pe)$ .

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#### Example 4

This example shows the reduction of axial dispersion (the spreading of a plug of miscible solution along the principal direction of the flow) in a mixing apparatus according to one embodiment of the present invention. Two channels having chevron-shaped undulations, each  $200\text{ }\mu\text{m} \times 70\text{ }\mu\text{m} \times 20\text{ cm}$ , were produced as shown schematically in FIG. 8. The top mixing channel had ten mixing cycles near the entrance while the bottom had mixing cycles substantially throughout its length. Steady streams of alkaline phosphatase (AP) and fluorocien diphosphate (FdP) were introduced into each mixing channel. AP reacted as in came in contact with FdP to produce a fluorescent molecule, fluorocien. The Péclet number was determined to be less than about  $1.7 \times 10^4$ .

The insets are copies of confocal images of the cross-section of the mixing channel. Specifically, the left insets are copies of confocal images after ten mixing cycles while the right insets, measured about 16 cm downstream, show the effect without (top) mixing and with (bottom) continuous mixing (at about 100 mixing cycles). As shown in the contrasting images, the fluid that is continuously mixed (bottom) was more homogeneous than the fluid that was not mixed (top). Homogeneity in these images indicates that the distribution of lifetimes (of the reaction product) in the flow is narrow and that there is little axial dispersion. Thus, this example demonstrates the benefit of using aspects of the present invention to increase conversion efficiency in a laminarly flowing reactive system.

#### Example 5

FIGS. 9a-b shows axial dispersion with and without efficient mixing and demonstrates the reduction of dispersion of a plug of miscible solution in a chaotically stirred Poiseuille flow (FIG. 9b) relative to an unstirred Poiseuille flow (FIG. 9a).

FIG. 9a shows unstirred Poiseuille flow in a rectangular channel that is  $21\text{ cm} \times 200 \times 70\text{ }\mu\text{m}^2$ . FIG. 9b shows stirred flow in a staggered herringbone mixing apparatus that is  $21\text{ cm} \times 200 \times 85\text{ }\mu\text{m}^2$ . A plug of fluorescent dye was introduced into both structures. The traces represent the time evolution of the total fluorescence intensity as observed with a fluorescence microscope having  $5\times$  lens that averages over the cross-section of the channel at positions 0.20 cm (100), 0.62 cm (102), 1.04 cm

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(104), 1.46 cm (106), and 1.88 cm (108) downstream from the entrance of the channel. These distances corresponded to 10, 30, 50, 70, and 90 mixing cycles, respectively. In the unstirred case, FIG. 9a, the plug was distorted and spread over most of the length of the channel. In the chaotically stirred flow, FIG. 9b, the plug retained its shape and  
 5 broadened only mildly. The appropriate fluid flow parameter were calculated to be  $U_a \sim 0.3$  cm/s;  $Pe \sim 1.5 \times 10^4$ ;  $L_{max}/h = 20$  cm/80  $\mu$ m = 2500.

FIG. 9a illustrated that for high  $Pe$ , the width of a plug in an unstirred Poiseuille flow grew linearly with time at the maximum flow speed,  $U_{max}$  (the fluid at the center of the channel moves at  $U_{max}$  while fluid at the walls is stationary); this rapid  
 10 broadening will continue for a distance down the channel,  $L \sim hPe$ . The traces record the total fluorescence intensity, integrated over the cross-section of the channel, as a function of time at equally spaced positions along the channel.

In the absence of stirring, the initial distribution of fluorescence rapidly distorted. The peak intensity also drastically reduced. The plugs developed long tails  
 15 due to the fluorescent solution that was trapped in the slowly moving regions of the flow near the walls. This effect, it is believed, is detrimental for the transfer of discrete plugs of fluid in laminar flows in channels and pipes.

In contrast, in chaotically stirred flow, shown in FIG. 9b, a plug of solution broadened more slowly because, it is believed, volumes of the solution moved between  
 20 fast and slow regions of the flow. Thus, the broadening of a plug should rapidly become diffusive, i.e., it is believed that the broadening is proportional to  $\sqrt{t}$  and should occur after  $n_m$  cycles with an effective diffusivity,  $D_{eff}$ , that is a function of the molecular diffusivity and the characteristics of the flow as discussed by Jones et al. in *J. Fluid Mech*, 280, pp. 149-172 (1994), which is incorporated by reference in its  
 25 entirety.

The traces shown in FIG. 9b demonstrated improved reduction of dispersion in a flow that was stirred in a mixing apparatus with a staggered herringbone structure, i.e., chaotically stirred flow. As shown in FIG. 9b, in the chaotically stirred flow, the shape of the distribution of fluorescence was largely maintained, and the peak intensity  
 30 dropped gradually.

Those skilled in the art would readily appreciate that all parameters and configurations described herein are meant to be exemplary and that actual parameters

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and configurations will depend upon the specific application for which the mixing systems and methods of the present invention are used. Those skilled in the art will recognize, or be able to ascertain using no more than routine experimentation, many equivalents to the specific embodiments of the invention described herein. For

5 example, those skilled in the art may recognize that the mixing apparatus of the present invention may be used to mix a fluid having a solid dissolving therein and that the present invention may be used to improve the transfer properties, heat or mass transfer, of a fluid flowing adjacent a surface having the features of the present invention. Moreover, the present invention can be seen to provide efficient mixing at low

10 Reynolds numbers but should be effective for any non-turbulent flow, Reynolds number less than about 2300, and need not be restricted to a systems with Reynolds number less than 100 or with dimensions less 1000  $\mu\text{m}$ .

It is, therefore, to be understood that the foregoing embodiments are presented by way of example only and that, within the scope of the appended claims and

15 equivalents thereto, the invention may be practiced otherwise than as specifically described. The present invention is directed to each individual feature, system, or method described herein. In addition, any combination of two or more such features, systems or methods, if such features, systems or methods are not mutually inconsistent, is included within the scope of the present invention.

20 What is claimed:

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**Claims**

1. An article comprising a microfluidic channel defined therein designed to have fluid flow therethrough in a principal direction, the microfluidic channel including a channel surface having at least one groove or protrusion defined therein, the at least one  
5 groove or protrusion having a first orientation that forms an angle relative to the principal direction.
2. The article of claim 1, wherein the microfluidic channel has at least one of a width and a depth that is less than about 1000  $\mu\text{m}$ .
- 10 3. The article of claim 2, wherein the microfluidic channel has at least one of a width and a depth that is less than about 500  $\mu\text{m}$ .
4. The article of claim 3, wherein the microfluidic channel has at least one of a  
15 width and a depth that is less than about 200  $\mu\text{m}$ .
5. The article of claim 1, wherein the substrate comprises a polymer.
6. The article of claim 1, wherein the angle is less than about 90 degrees.
- 20 7. The article of claim 1, wherein the groove or protrusion has a depth that is less than a width of the microfluidic channel.
8. The article of claim 1, wherein the groove or protrusion has a depth that is less  
25 than a depth of the microfluidic channel.
9. The article of claim 1, wherein the groove or protrusion has a width that is less than a width of the microfluidic channel.
- 30 10. The article of claim 1, wherein the microfluidic channel includes a first inlet.

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11. The article of claim 10, wherein the microfluidic channel includes a second inlet.
12. The article of claim 1, wherein the microfluidic channel has a substantially  
5 circular cross-section.
13. The article of claim 1, comprising a plurality of grooves or protrusions formed in the channel surface.
- 10 14. The article of claim 13, wherein each of the grooves or protrusions is parallel to each other.
- 15 15. The article of claim 14, wherein the parallel grooves or protrusions are periodically spaced along the channel surface to form a first set of parallel grooves or protrusions.
16. The article of claim 15, wherein the microfluidic channel has a width and the first set of parallel periodically-spaced grooves or protrusions traverse the width.
- 20 17. The article of claim 13, wherein the channel surface has a second set of parallel periodically-spaced grooves or protrusions traversing at least a portion of the channel surface at a second orientation.
- 25 18. The article of claim 17, wherein the second set of parallel periodically-spaced grooves or protrusions are at least partially coextensive with the first set of parallel periodically-spaced grooves or protrusions.
19. The article of claim 17, wherein the first and second sets of parallel grooves or protrusions form a repeating pattern.
- 30 20. The article of claim 1, wherein at least one groove or protrusion has at least two sections.

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21. The article of claim 20, wherein at least one section is substantially linear.
22. The article of claim 21, wherein the sections intersect to form at least one  
5 chevron-shaped groove.
23. The article of claim 22, wherein a plurality of chevron-shaped grooves or protrusions are formed in the channel surface.
- 10 24. The article of claim 23, wherein the chevron-shaped grooves or protrusions are periodically spaced along the channel surface.
25. The article of claim 1, wherein a second groove or protrusion is defined in the channel surface, the second groove or protrusion having a second orientation relative to  
15 the principal direction.
26. The article of claim 1, wherein the substrate has a network of microfluidic channels fluidly connected to the microfluidic channel.
- 20 27. The article of claim 1, wherein the microfluidic channel is formed in a unitary substrate.
28. An article comprising a microfluidic channel constructed and arranged to have a fluid flowing therethrough while creating a transverse flow component in the fluid.  
25
29. The article of claim 28, wherein the microfluidic channel is constructed and arranged so that fluid flowing therethrough has a Reynolds number that is less than about 12.
- 30 30. The article of claim 29, wherein the microfluidic channel is constructed and arranged so that fluid flowing therethrough has a Reynolds number that is less than about 5.

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31. The article of claim 28, wherein the microfluidic channel has a width that is less than about 1000  $\mu\text{m}$ .

5 32. The article of claim 28, further comprising a network of microfluidic channels fluidly connected to the microfluidic channel.

33. The article of claim 28, wherein the microfluidic channel is constructed and arranged to create at least one helical flow path in a fluid flowing therethrough.

10

34. The article of claim 28, wherein the microfluidic channel is constructed and arranged to have a substantially circular cross-section.

15 35. The article of claim 28, wherein the microfluidic channel is constructed and arranged to have a rectangular cross-section.

36. The article of claim 28, wherein the transverse flow component is created regardless of the Reynolds number of the fluid flowing in the microfluidic channel.

20 37. An article comprising a structure having a channel defined therein, the channel designed to have a fluid flowing therethrough in a principal direction, the channel including a channel surface having a plurality of chevron-shaped grooves or protrusions formed in at least a portion of the channel surface so that each chevron-shaped groove or protrusion has an apex that defines an angle.

25

38. The article of claim 37, wherein the angle of the apex is about 45-degrees.

39. The article of claim 37, wherein the channel includes a first set of chevron-shaped grooves or protrusions and a second set of chevron-shaped grooves or  
30 protrusions.



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40. The article of claim 39, wherein the apex of each of the first set of chevron-shaped grooves or protrusions are aligned offset relative to the apex of each of the second set of chevron-shaped grooves or protrusions.
- 5 41. The article of claim 40, wherein the structure comprises a capillary tube.
42. The article of claim 40, wherein the structure comprises a polymer.
43. The article of claim 37, wherein the channel has a width that is less than about  
10 1000  $\mu\text{m}$ .
44. The article of claim 43, wherein the channel has a width that is less than about 200  $\mu\text{m}$ .
- 15 45. The article of claim 37, wherein the channel is fluidly connected to a network of microfluidic channels.
46. The article of claim 37, wherein the chevron-shaped grooves or protrusions are periodically-spaced from each other.
- 20 47. The article of claim 37, wherein the channel has a rectangular cross-section.
48. The article of claim 37, wherein the channel has a circular cross-section.
- 25 49. The article of claim 37, wherein the channel is a microfluidic channel.
50. The article of claim 37, wherein the channel is defined on a unitary structure.
51. A structure comprising:  
30 a first channel having a width that is less than about 1000  $\mu\text{m}$ ;  
a second channel having a width that is less than about 1000  $\mu\text{m}$ ; and

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a third channel having a principal direction and a width that is less than about 1000  $\mu\text{m}$ , the third channel connecting the first and second channels and comprising channel surfaces having grooves or protrusions defined therein, the grooves or protrusions oriented at an angle relative to the principal direction.

5

52. The structure of claim 51, wherein the structure comprises a polymer.

53. A method for dispersing a material in a fluid comprising:

providing an article having a channel designed to have fluid flow therethrough  
10 in a principal direction, the channel including a channel surface having at least one groove or protrusion therein that traverses at least a portion of the channel surface, at least one groove or protrusion oriented at an angle relative to the principal direction;  
and

causing the fluid in the channel to flow laminarily along the principal direction.

15

54. The method of claim 53, wherein the fluid flowing in the channel has a Reynolds number that is less than about 100.

55. The method of claim 54, wherein the fluid flowing in the channel has a  
20 Reynolds number that is less than about 10.

56. The method of claim 55, wherein the fluid flowing in the channel has a Reynolds number that is less than about 5.

25 57. The method of claim 53, wherein the step of causing the fluid to flow in the channel results in a fluid residence time in the channel of less than about 20 seconds.

58. A method comprising:

causing a first fluid to flow in a channel at a Reynolds number that is less than  
30 about 100;

causing a second fluid to flow in the channel at a Reynolds number that is less than about 100; and

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creating a transverse flow component in the first and the second fluids to promote mixing between the first and second fluids.

59. The method of claim 58, wherein the channel has a width that is less than about  
5 1000  $\mu\text{m}$ .

60. The method of claim 59, wherein the step of creating a transverse flow component creates at least one helical flow path.

10 61. The method of claim 58, wherein the second fluid has a Reynolds number that is about equal to the Reynolds number of the first fluid.

62. The method of claim 61, wherein the first fluid has a composition that differs from a composition of the second fluid.

15

63. A method for forming a microfluidic article comprising:  
forming a first topological feature that has a smallest dimension that is less than about 1000  $\mu\text{m}$  on a surface of a mold substrate;  
forming a second topological feature on the first topological feature to form a  
20 mold master, the second topological feature characterized by a length that traverses at least a portion of a section of the first topological feature;  
placing a hardenable material on the surface;  
hardening the material thereby creating a molded article having a microfluidic channel shaped from the first topological feature and at least one groove or protrusion  
25 shaped from the second topological feature; and  
removing the microfluidic article from the mold master.

64. The method of claim 63, wherein the hardenable material comprises a cross-linkable polymer.

30

65. The method of claim 64, wherein the step of hardening the material comprises applying heat to the material.

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66. The method of claim 65, wherein the groove or protrusion has a depth that is less than a width of the first topological feature.

5 67. A method for producing a helical flow path in a fluid flowing along a principal direction comprising:

providing a structure having a surface with a plurality of substantially linear grooves or protrusions oriented at an angle relative to the principal direction, the grooves or protrusions formed to be parallel to and periodically spaced from each other;

10 and

causing the fluid to flow along the surface, the fluid flowing adjacent the surface having a Reynolds number that is less than about 100.

1/5

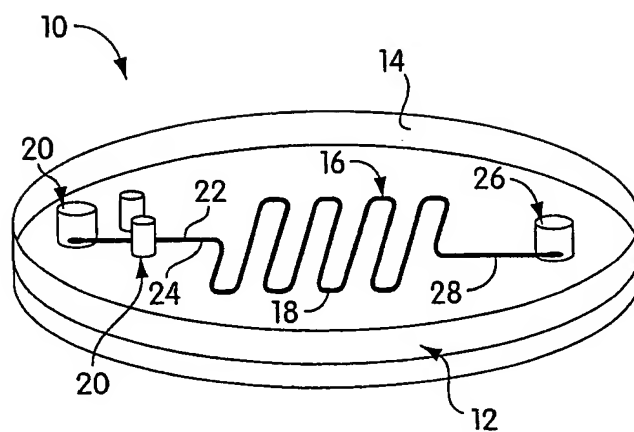


Fig. 1

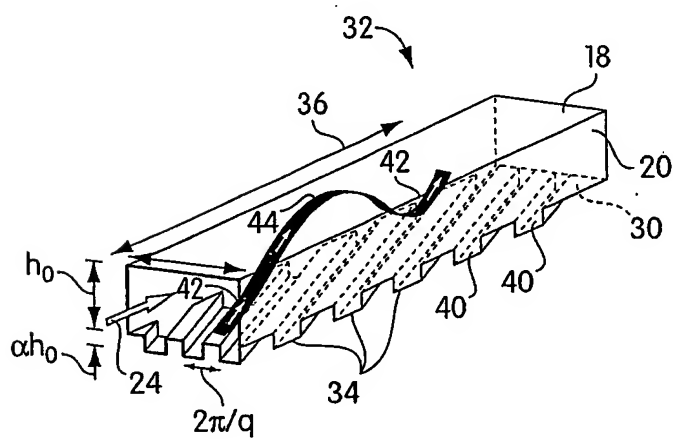


Fig. 2a

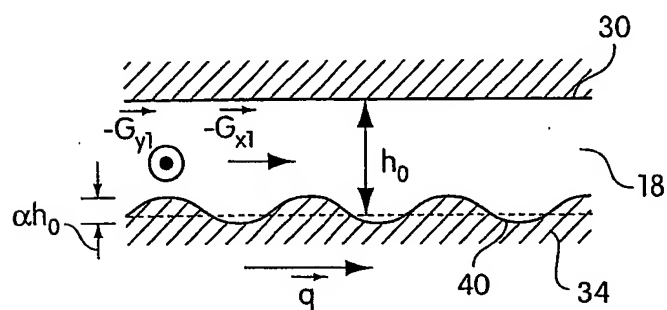


Fig. 2b

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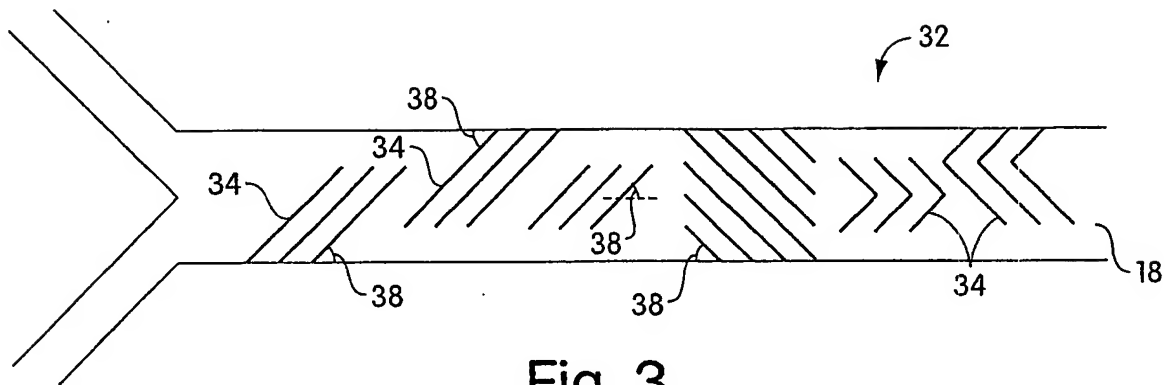


Fig. 3

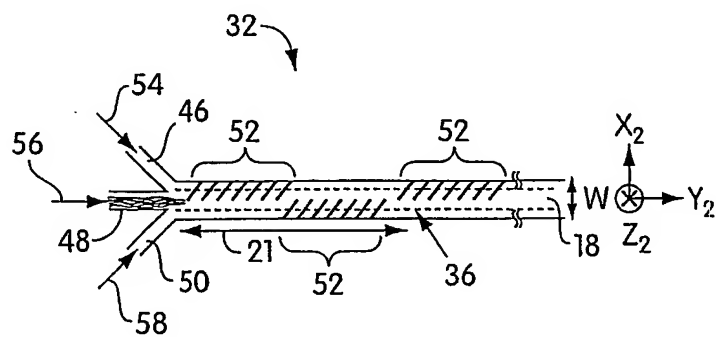


Fig. 4a

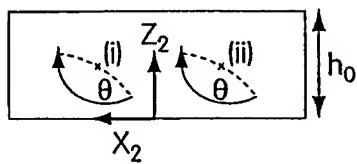


Fig. 4b

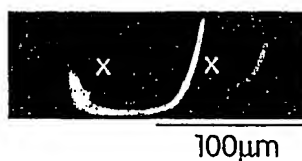


Fig. 4c

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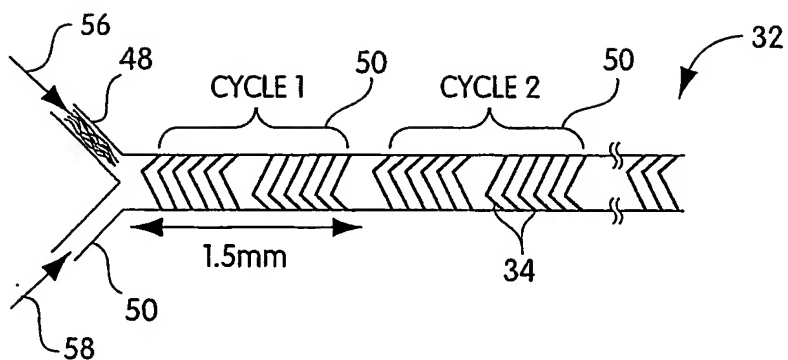


Fig. 5

100μm



Fig. 6a



Fig. 6b



Fig. 6c



Fig. 6d

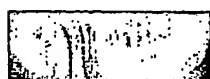


Fig. 6e

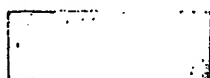


Fig. 6f

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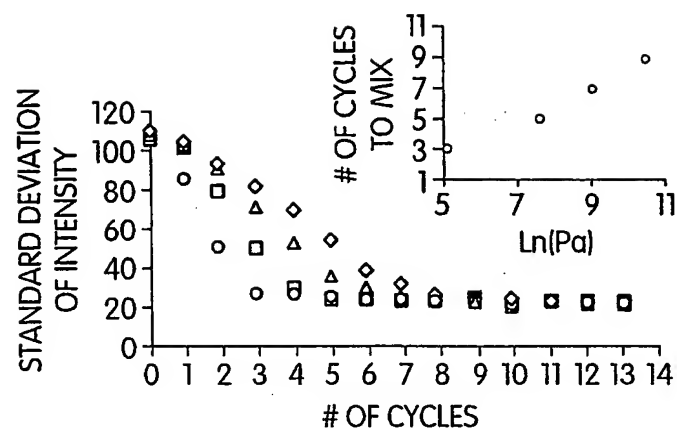


Fig. 7

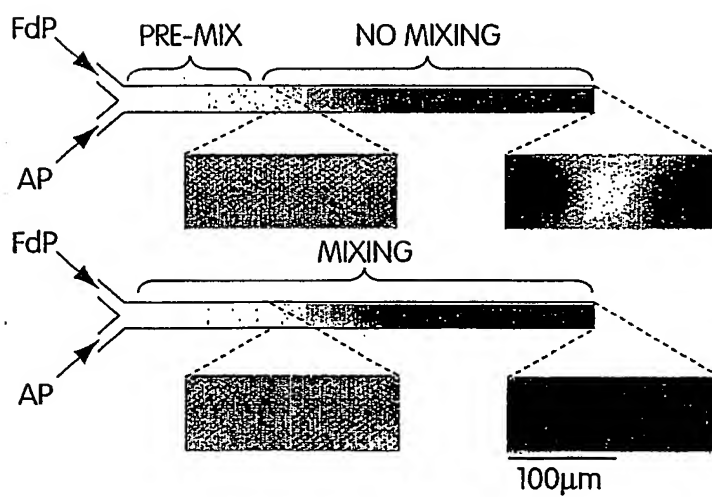


Fig. 8



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UNSTIRRED:

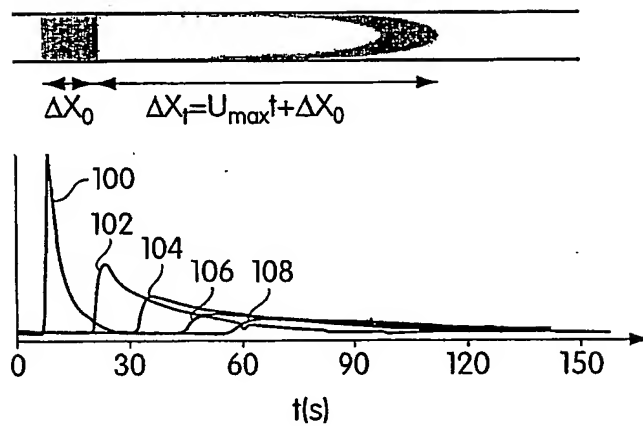


Fig. 9a

CHAOTICALLY STIRRED:

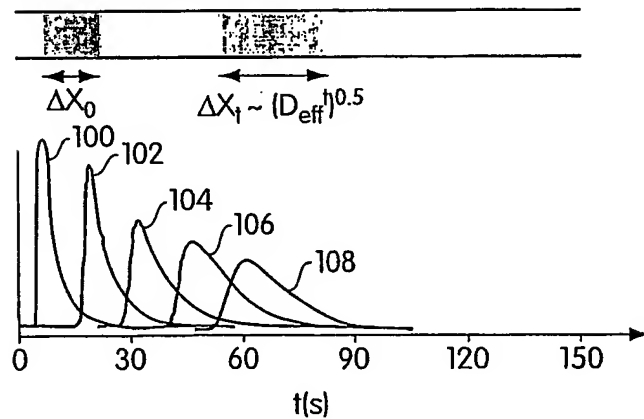


Fig. 9b